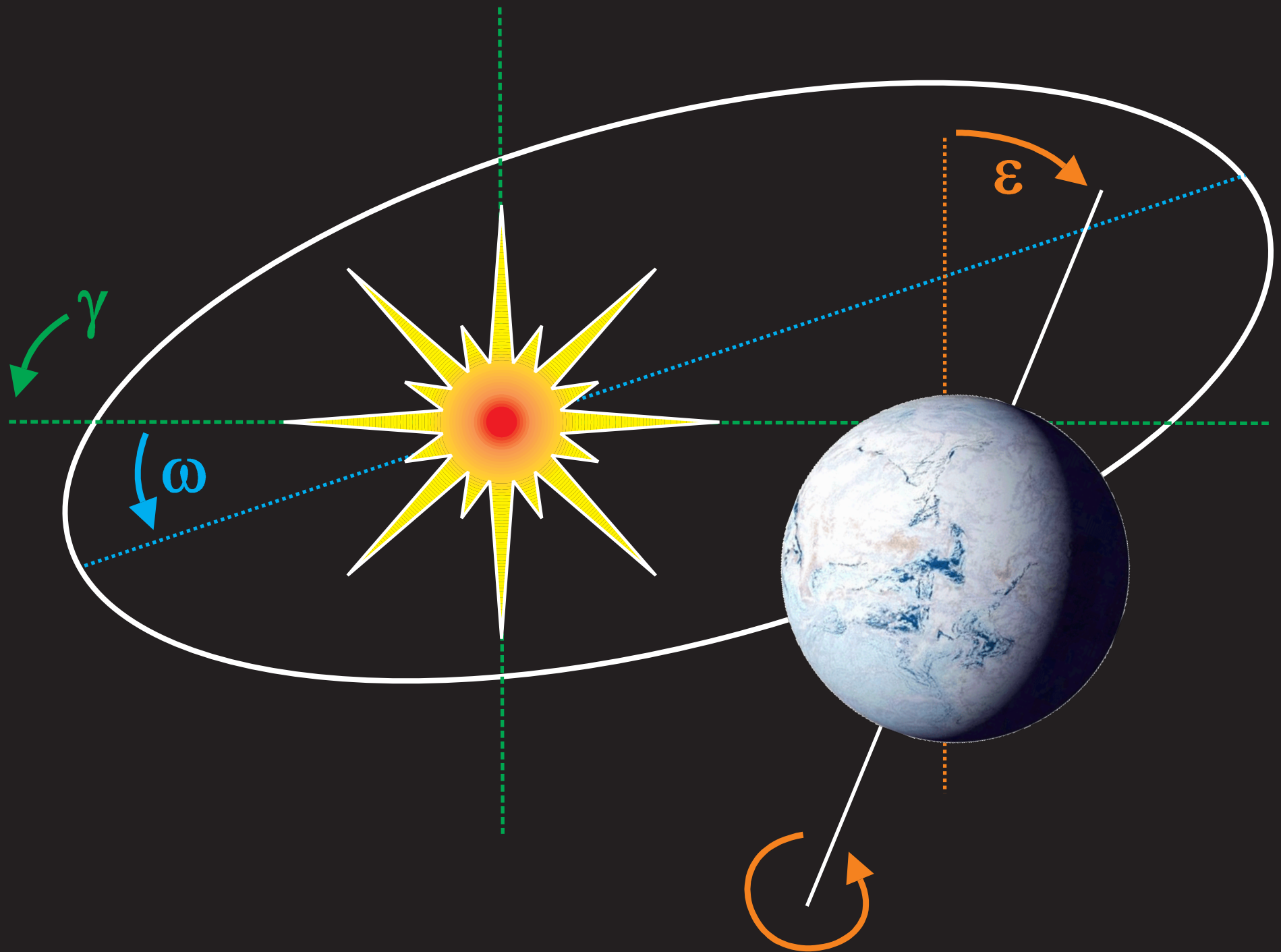


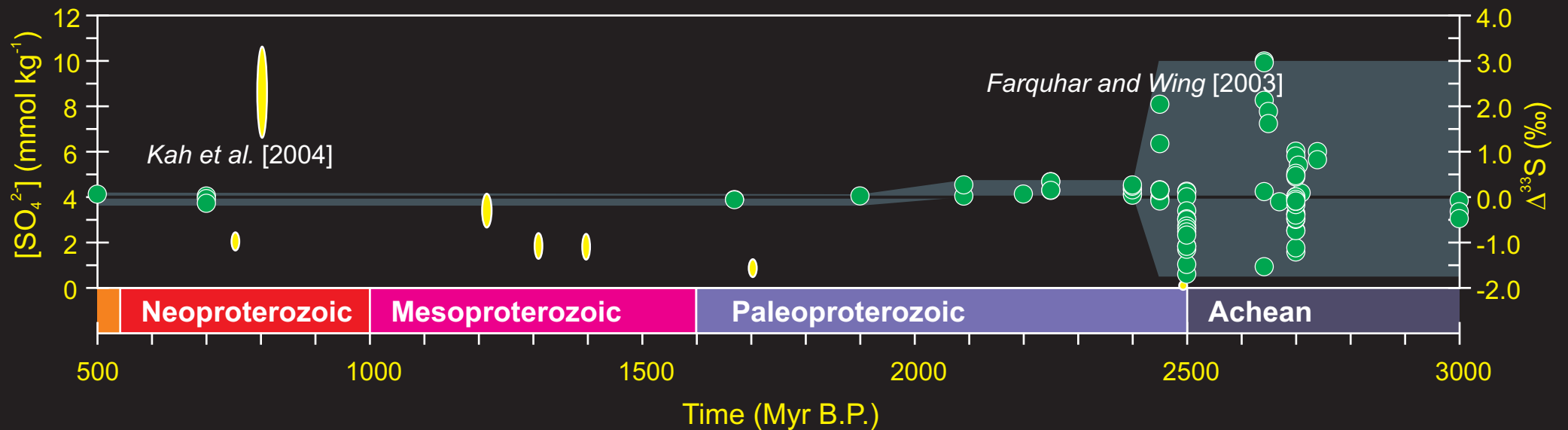
Snowball Earth



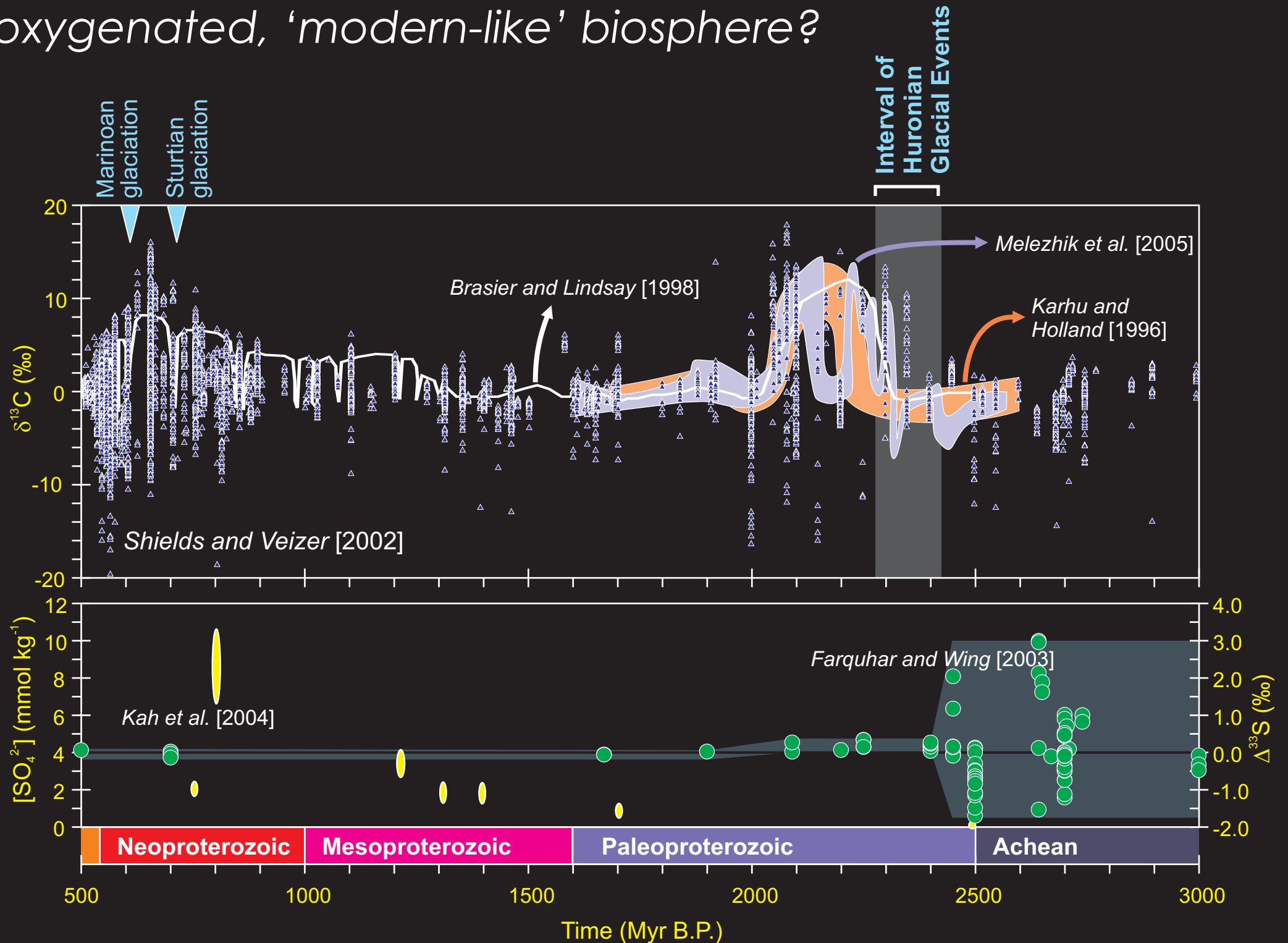
The Neoproterozoic: Gateway to a metazoan-dominated, oxygenated, 'modern-like' biosphere?



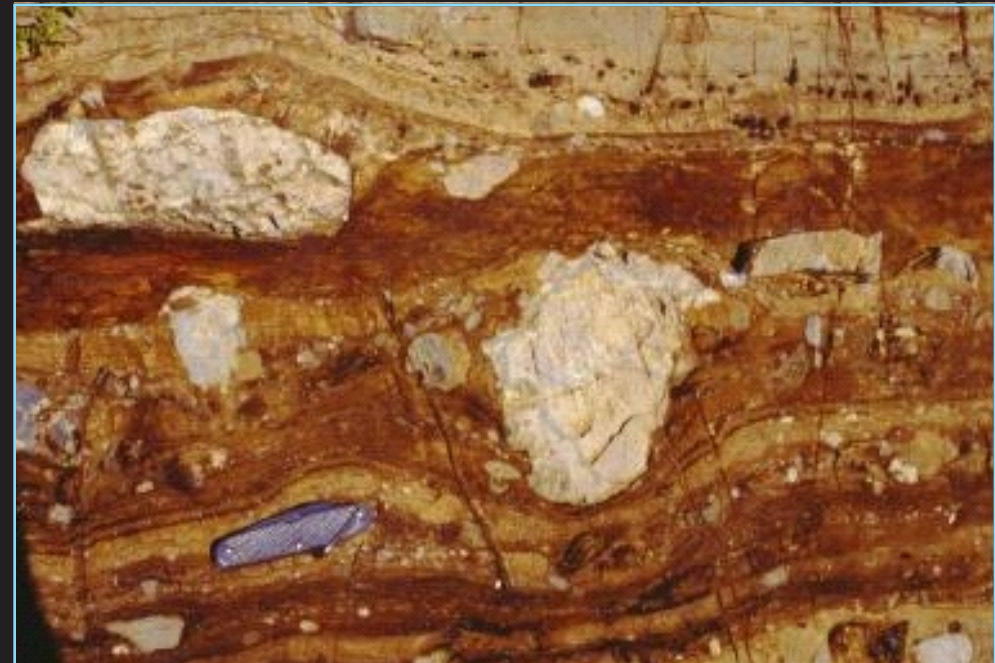
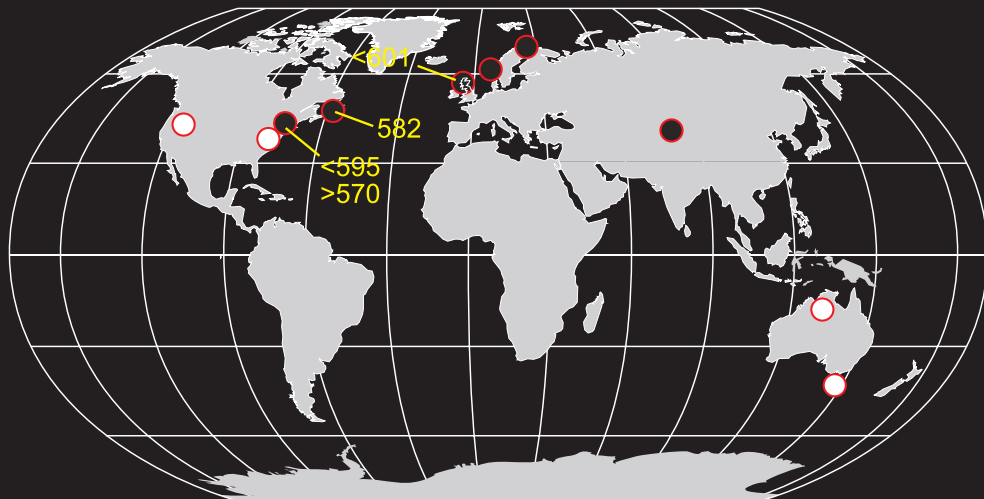
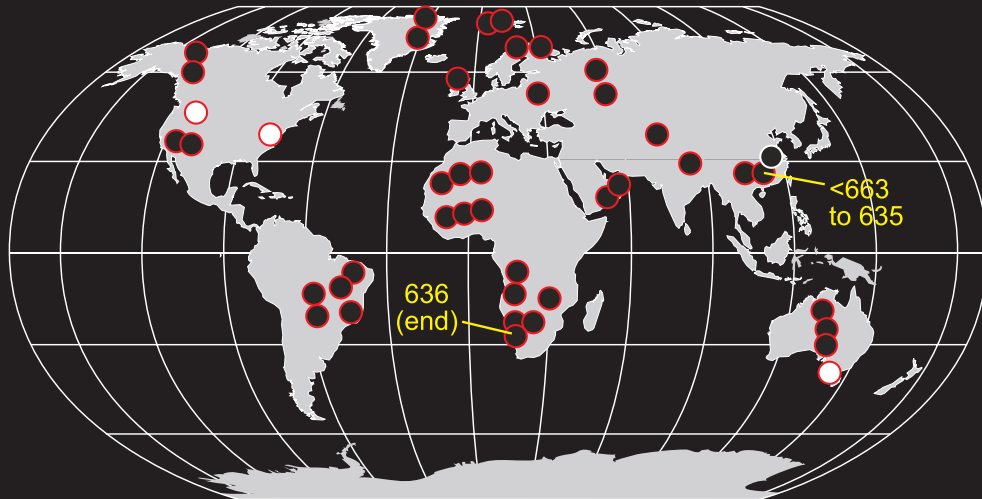
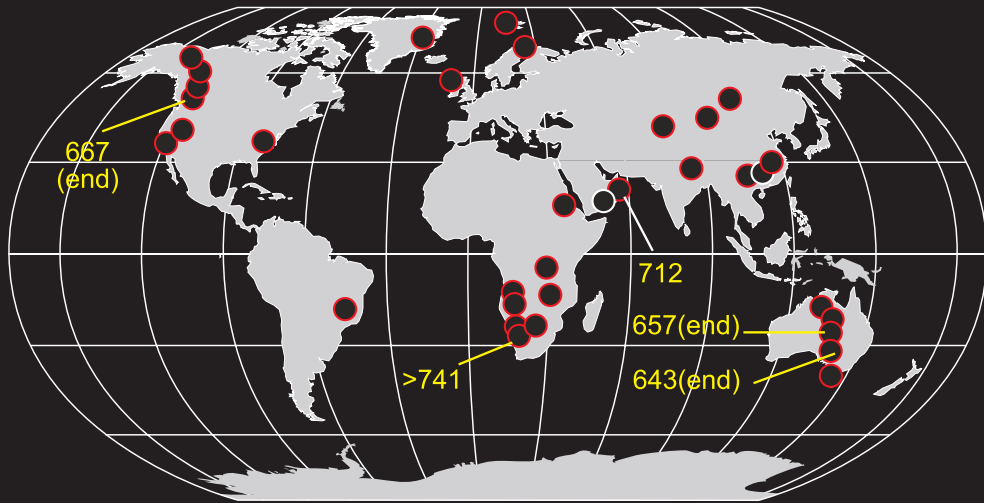
The Neoproterozoic: Gateway to a metazoan-dominated, oxygenated, 'modern-like' biosphere?



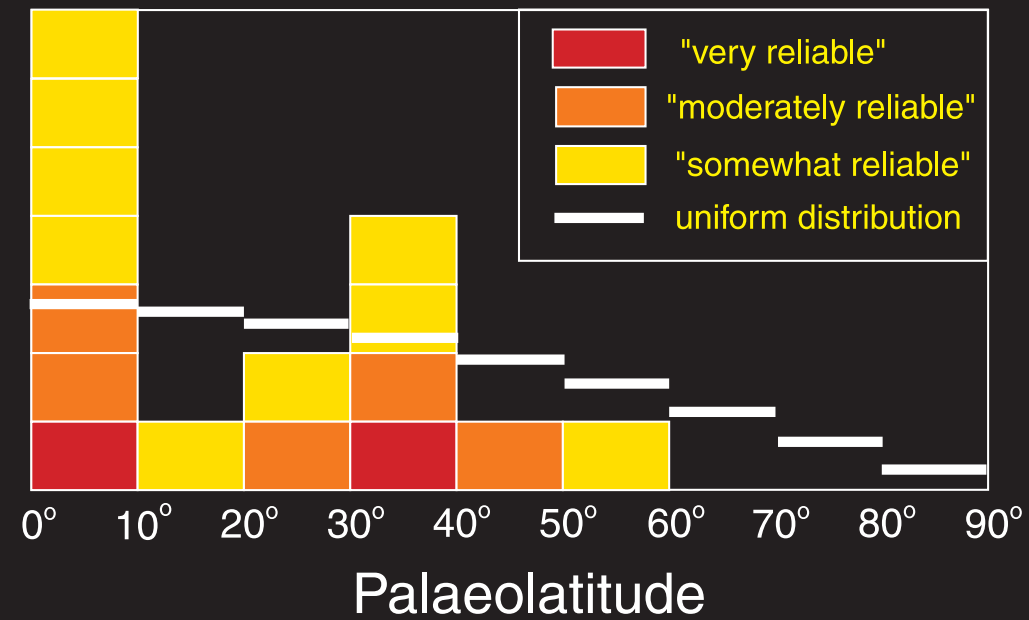
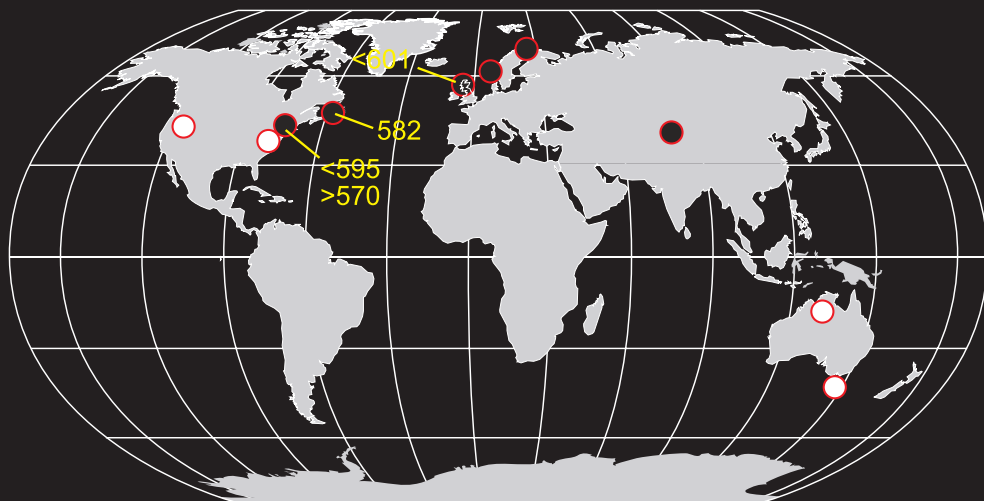
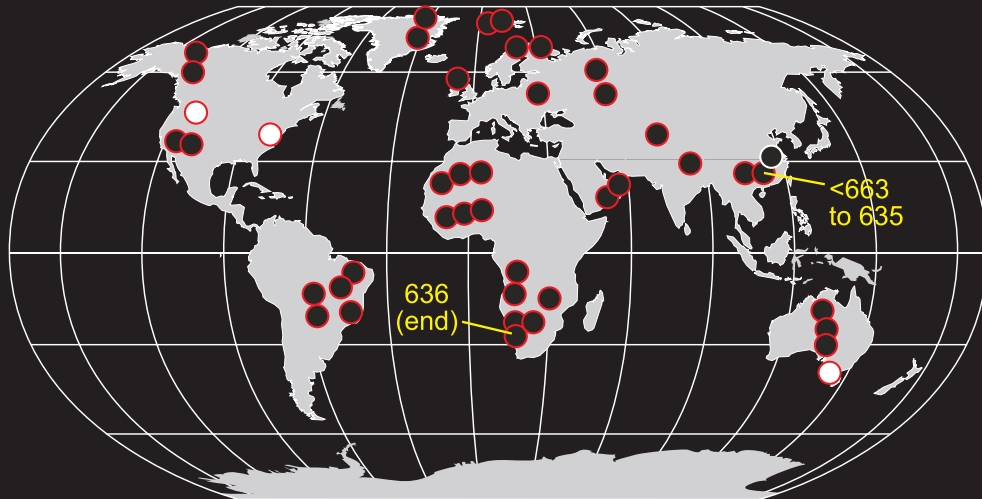
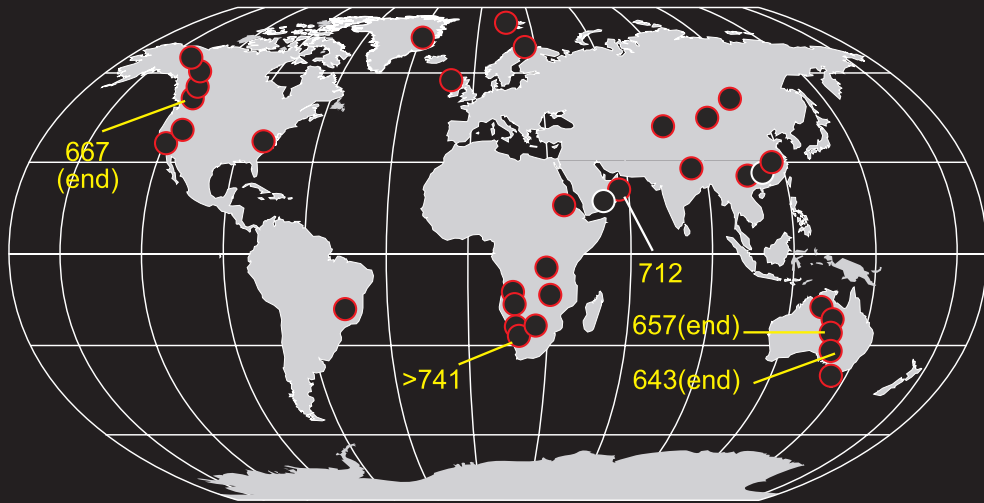
The Neoproterozoic: Gateway to a metazoan-dominated, oxygenated, 'modern-like' biosphere?



Evidence for glaciation



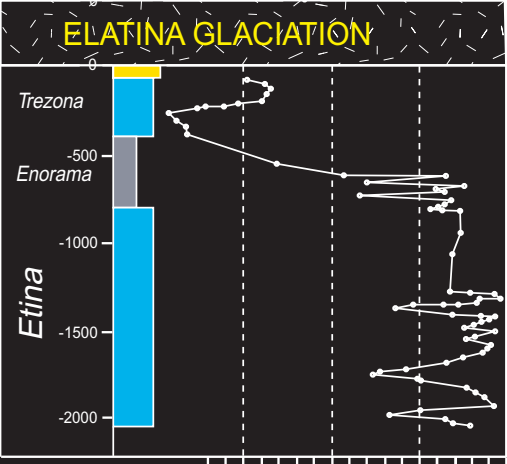
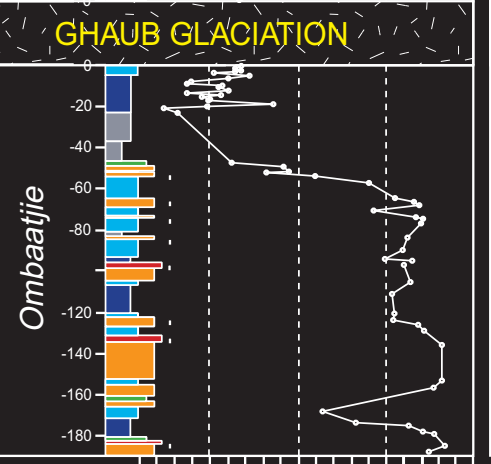
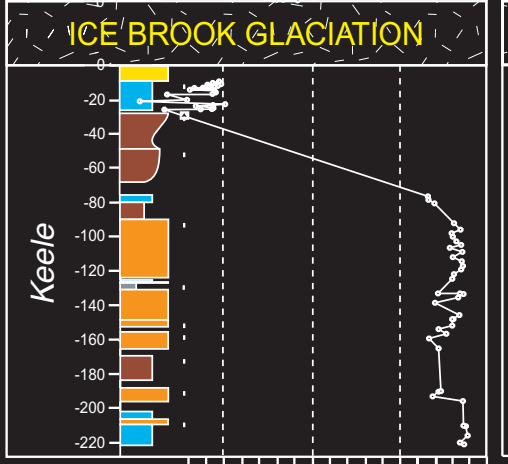
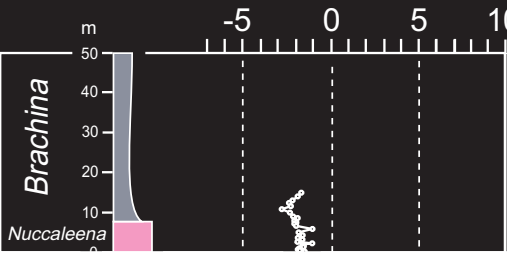
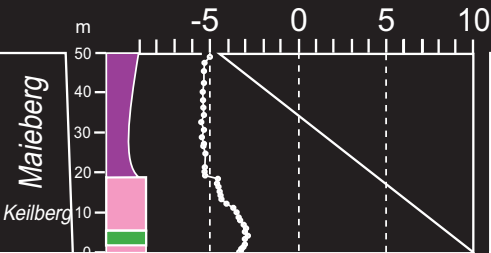
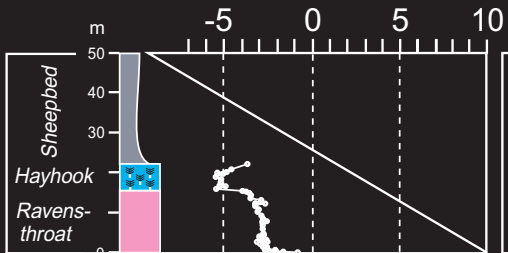
Evidence for glaciation



CANADA

NAMIBIA

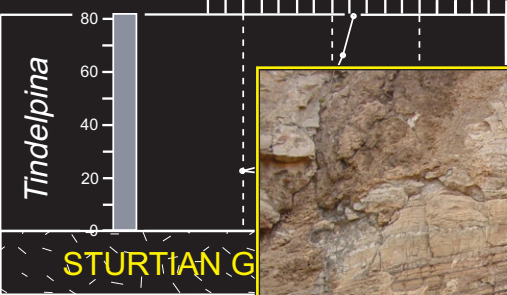
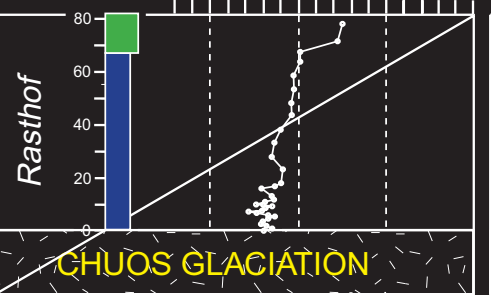
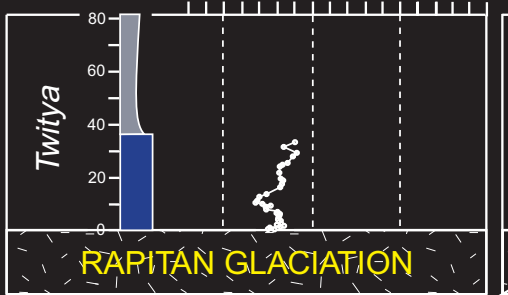
AUSTRALIA



SECTION NOT SHOWN

SECTION NOT SHOWN

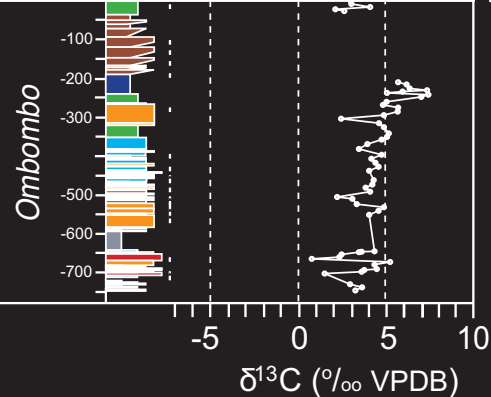
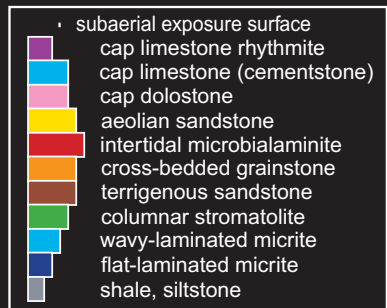
SECTION NOT SHOWN



RAPITAN GLACIATION

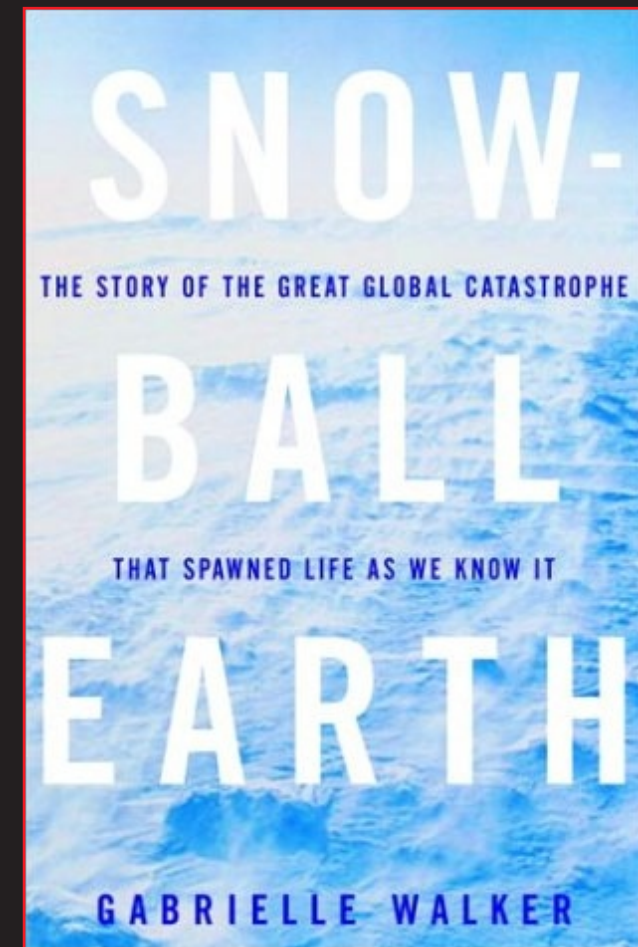
CHUOS GLACIATION

STURTIAN G



'snowball Earth'

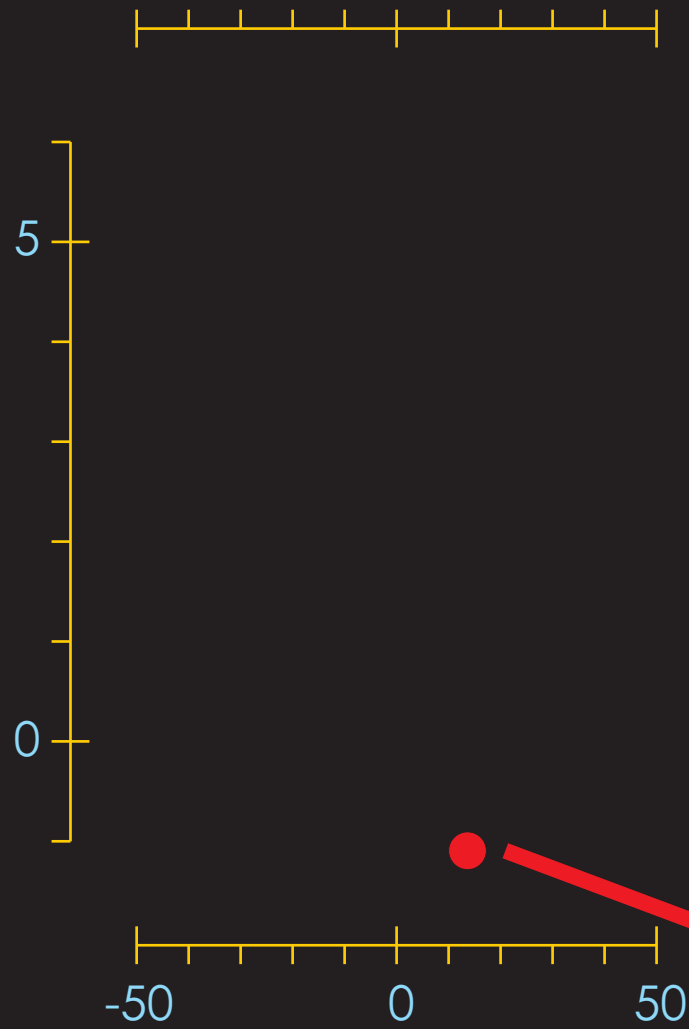
Hoffman et al. [1998] (*Science* **281**)



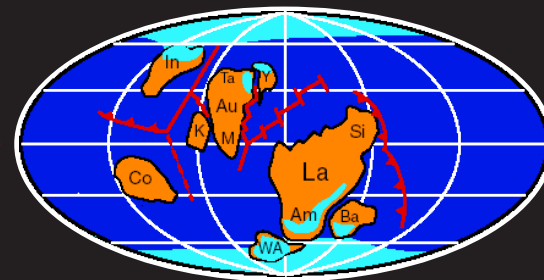
The snowball Earth hypothesis

[Hoffman and Schrag, 2002] (*Terra Nova* **14**, 129-155)

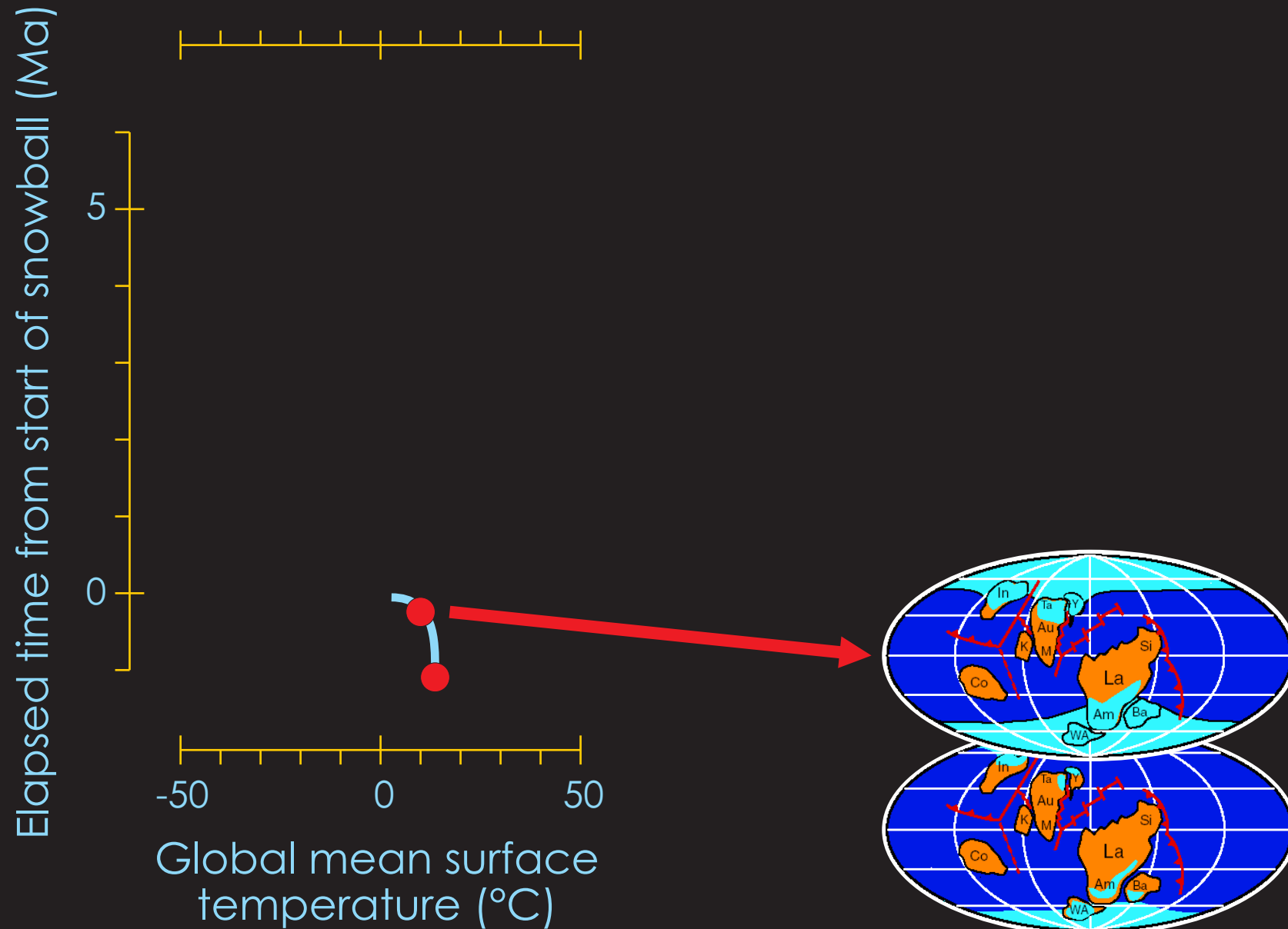
Elapsed time from start of snowball (Ma)



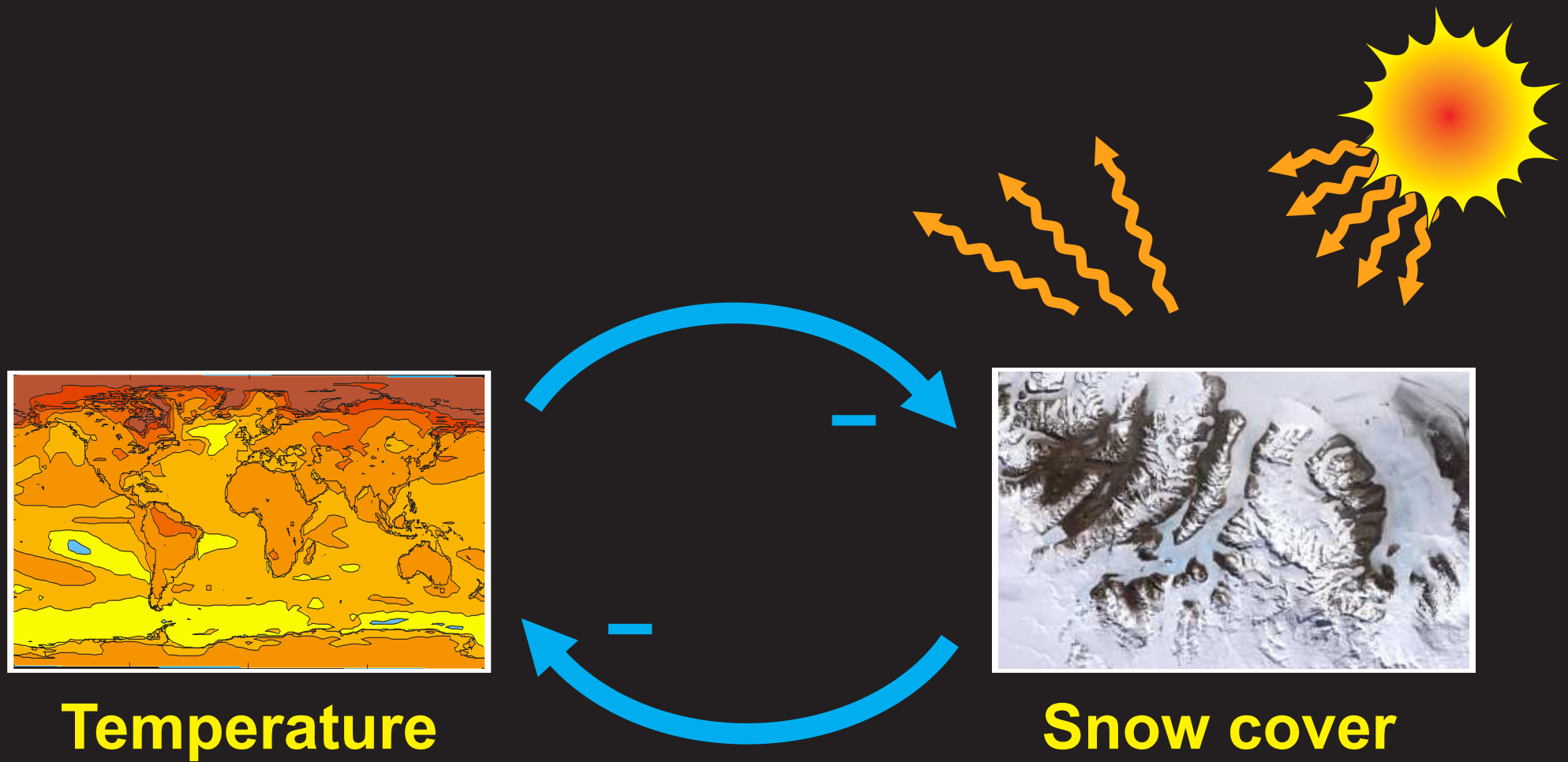
Global mean surface temperature (°C)



The snowball Earth hypothesis

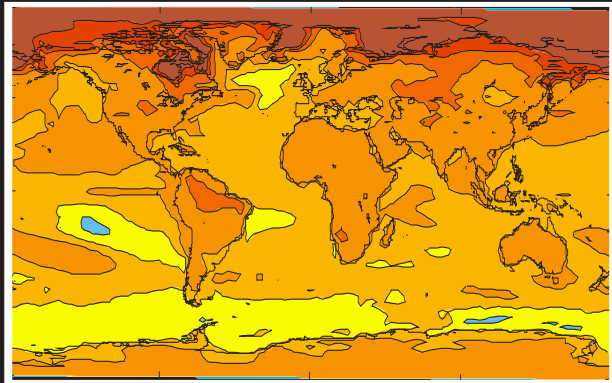
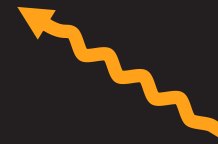
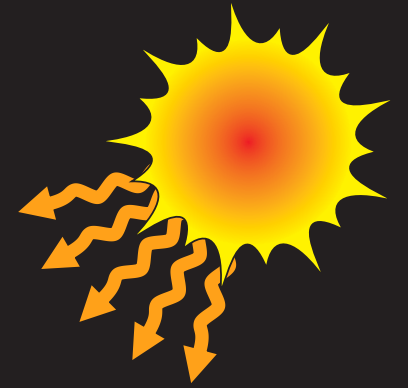


'Feedbacks'



positive "ice-albedo" feedback

'Feedbacks'



Temperature

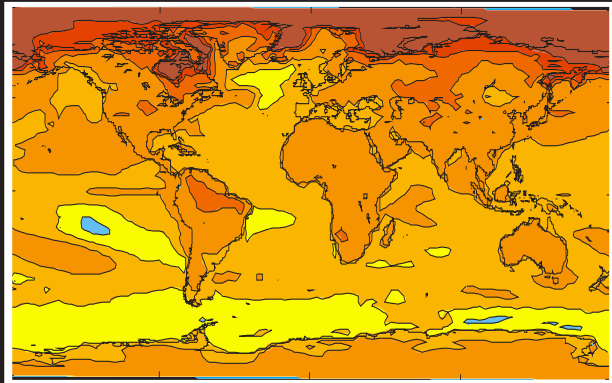


Snow cover



= $-1/2^{\circ}\text{C}$

'Feedbacks'

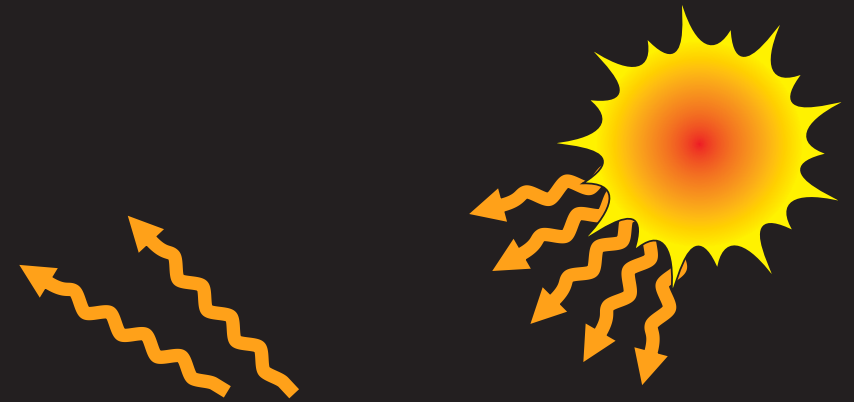


Temperature

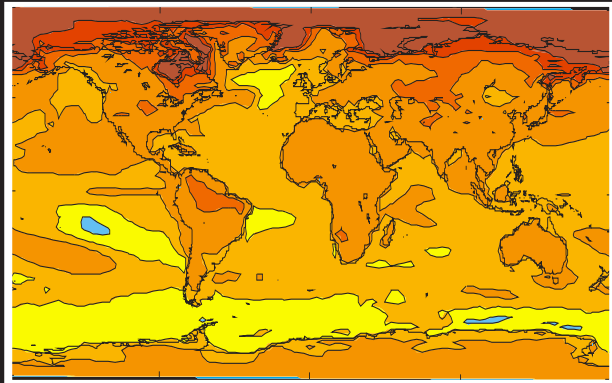


Snow cover

TOTAL CHANGE = $-1/2^{\circ}\text{C}$



'Feedbacks'



Temperature

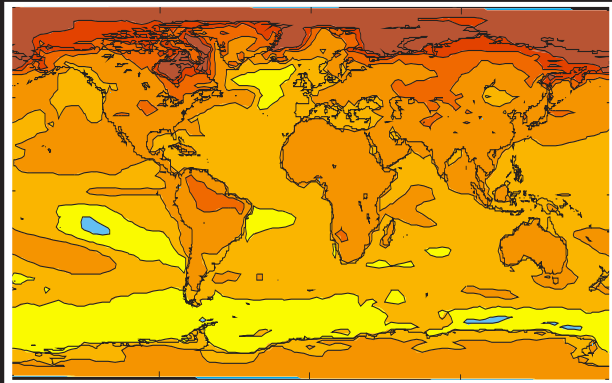


Snow cover



TOTAL CHANGE = $-1/2^{\circ}\text{C}$ - $1/4^{\circ}\text{C}$

'Feedbacks'



Temperature

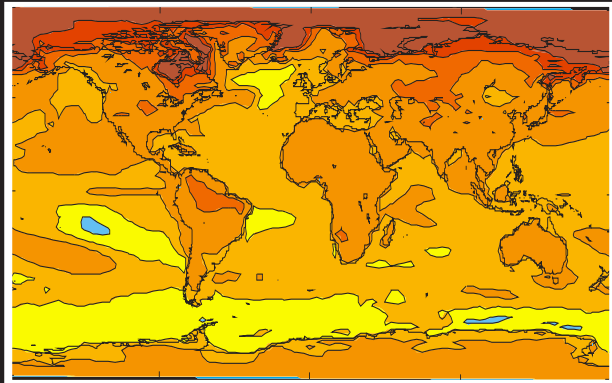
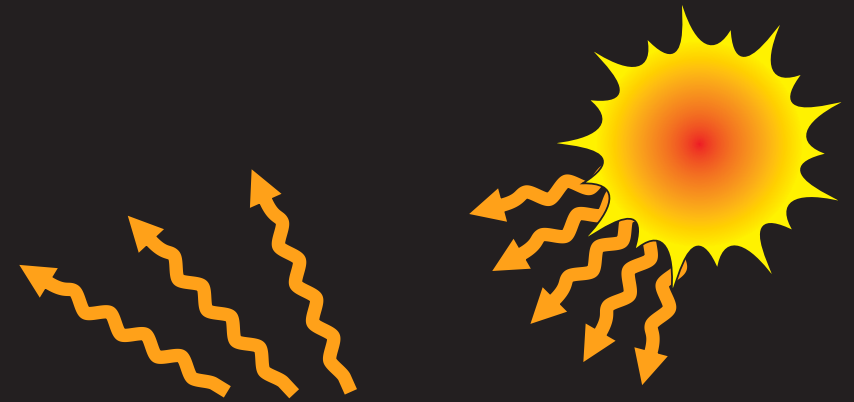


Snow cover

TOTAL CHANGE = $-1/2^{\circ}\text{C}$ - $1/4^{\circ}\text{C}$



'Feedbacks'



Temperature

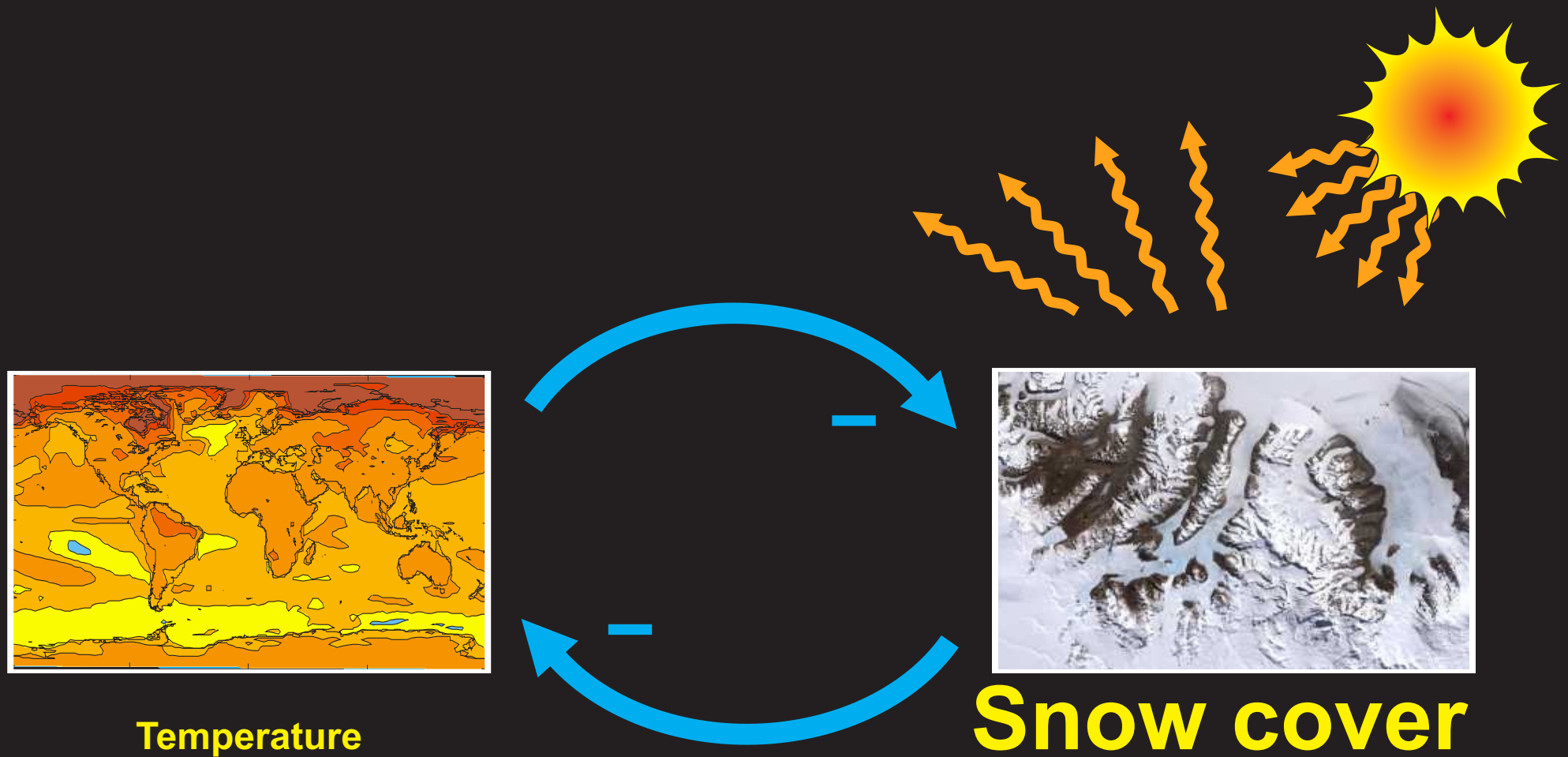


Snow cover



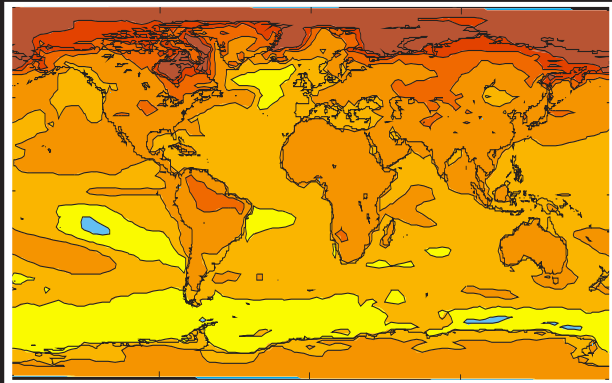
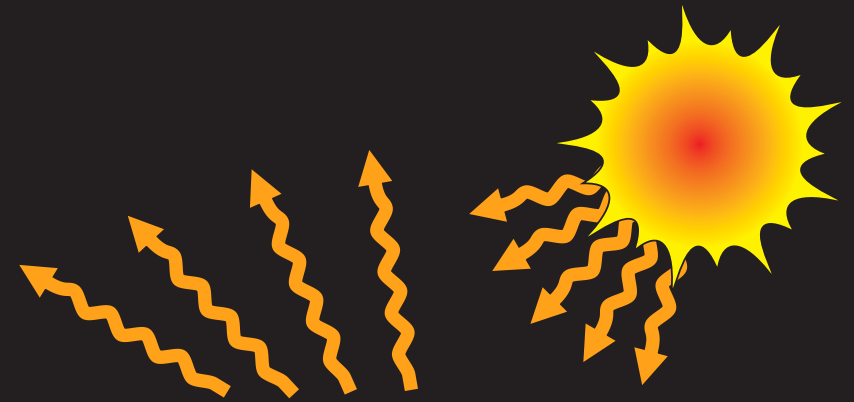
$$\text{TOTAL CHANGE} = -1/2^{\circ}\text{C} - 1/4^{\circ}\text{C} - 1/8^{\circ}\text{C}$$

'Feedbacks'



$$\text{TOTAL CHANGE} = -1/2^{\circ}\text{C} - 1/4^{\circ}\text{C} - 1/8^{\circ}\text{C} - 1/16^{\circ}\text{C} - \dots$$

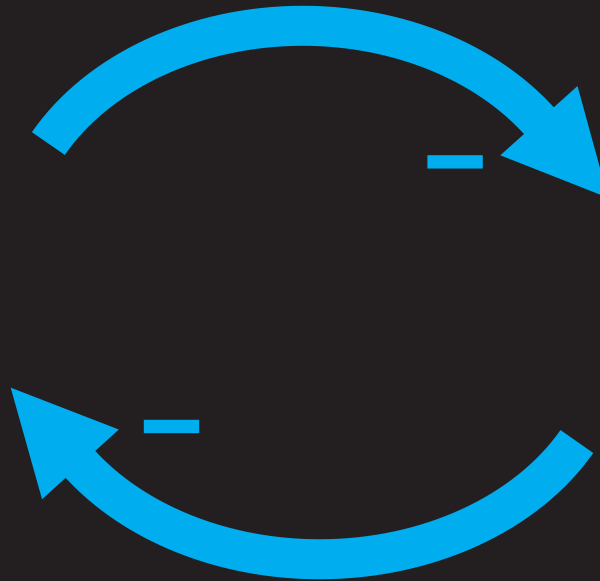
'Feedbacks' ('runaway')



Temperature

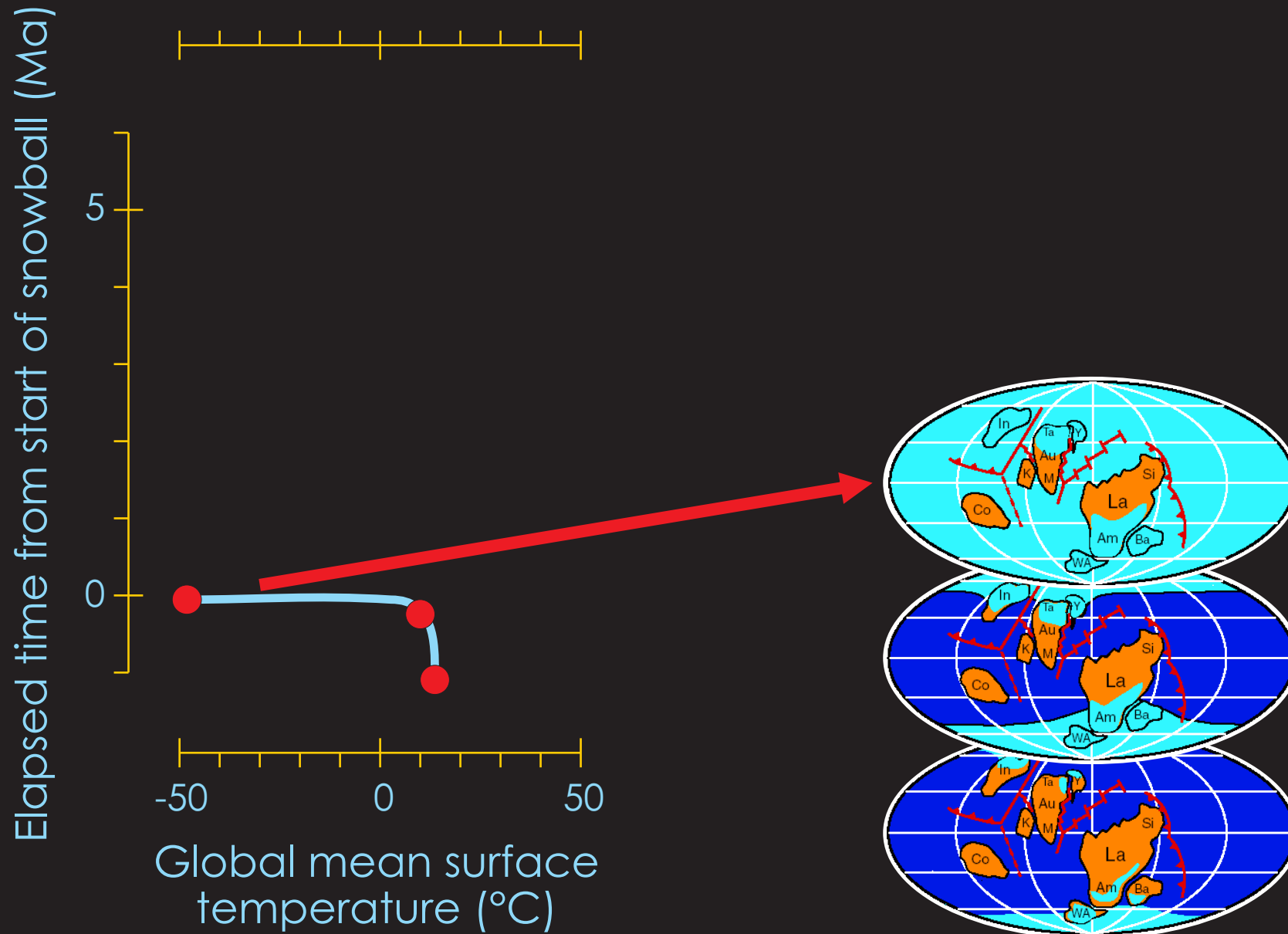


Snow cover

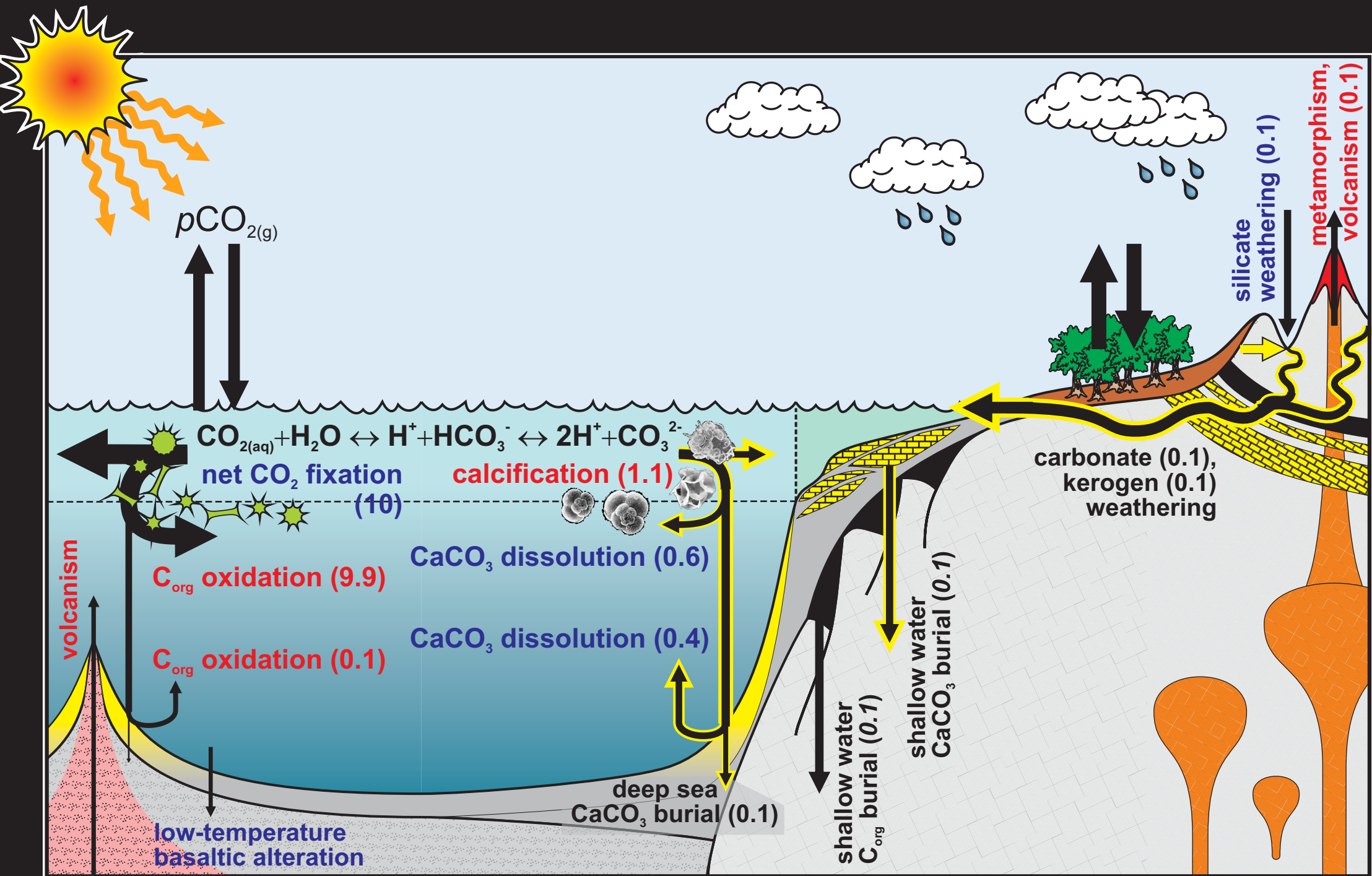


TOTAL CHANGE = $-1^{\circ}\text{C} - 2^{\circ}\text{C} - 4^{\circ}\text{C} - 8^{\circ}\text{C}$
-

The snowball Earth hypothesis



The global carbon cycle (modern)



Long-term controls on atmospheric $p\text{CO}_2$

Terrestrial weathering can be (approximately equally) divided into carbonate (CaCO_3) and calcium-silicate (' CaSiO_3 ') weathering:



Ultimately, the (alkalinity: Ca^{2+}) weathering products must be removed through carbonate precipitation and burial in marine sediments:



It can be seen that in (2) + (3), that the CO_2 removed (from the atmosphere) during weathering, is returned upon carbonate precipitation (and burial). In (1) + (3) (silicate weathering) CO_2 is permanently removed to the geological reservoir. This CO_2 must be balanced by mantle (/volcanic) out-gassing on the very long term.

Long-term controls on atmospheric $p\text{CO}_2$

Terrestrial weathering can be (approximately equally) divided into carbonate (CaCO_3) and calcium-silicate (' CaSiO_3 ') weathering:



Ultimately, the (alkalinity: Ca^{2+}) weathering products must be removed through carbonate precipitation and burial in marine sediments:



It can be seen that in (2) + (3), that the CO_2 removed (from the atmosphere) during weathering, is returned upon carbonate precipitation (and burial). In (1) + (3) (silicate weathering) CO_2 is permanently removed to the geological reservoir. This CO_2 must be balanced by mantle (/volcanic) out-gassing on the very long term.

Long-term controls on atmospheric $p\text{CO}_2$

Terrestrial weathering can be (approximately equally) divided into carbonate (CaCO_3) and calcium-silicate (' CaSiO_3 ') weathering:

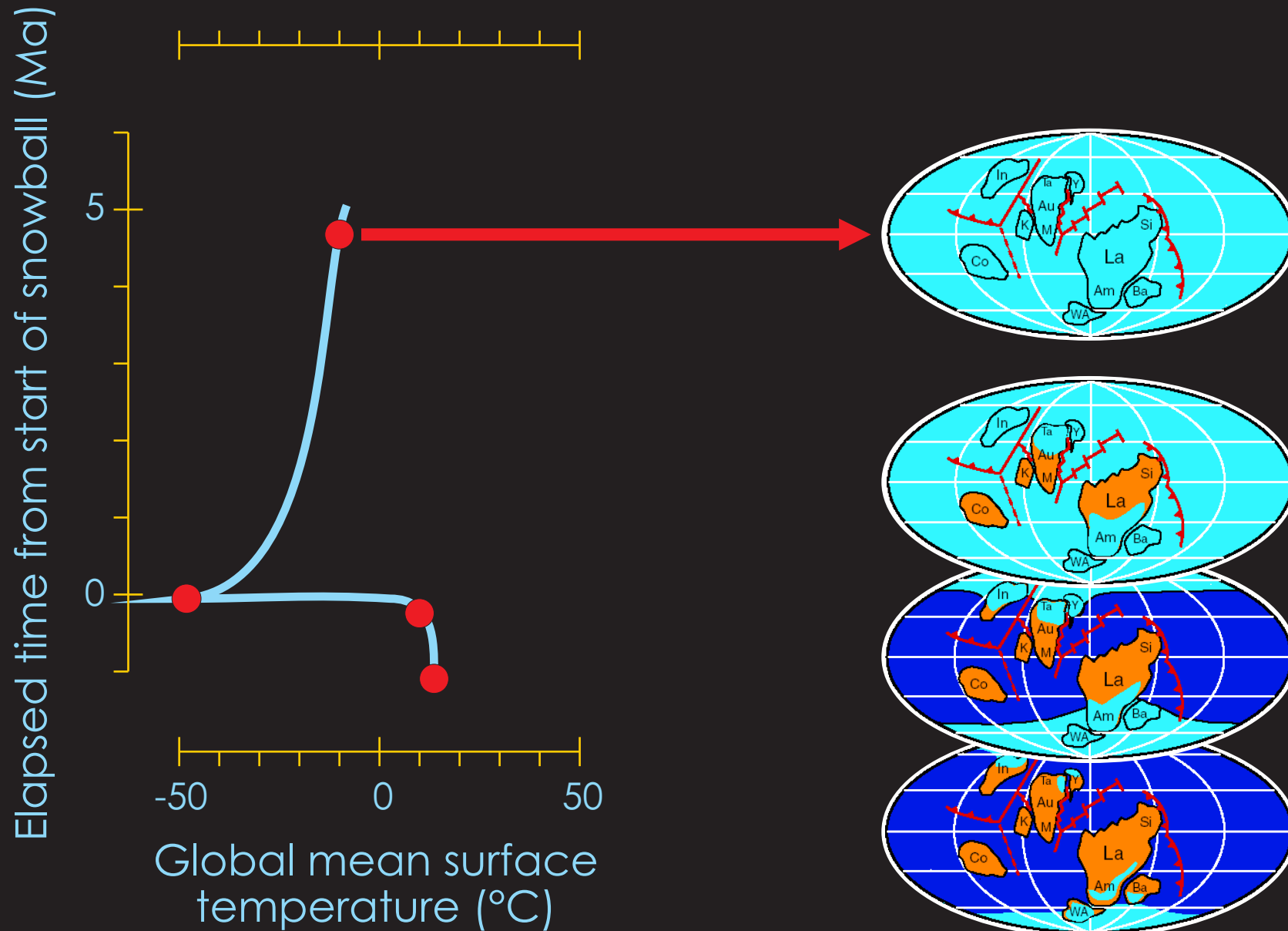


Ultimately, the (alkalinity: Ca^{2+}) weathering products must be removed through carbonate precipitation and burial in marine sediments:



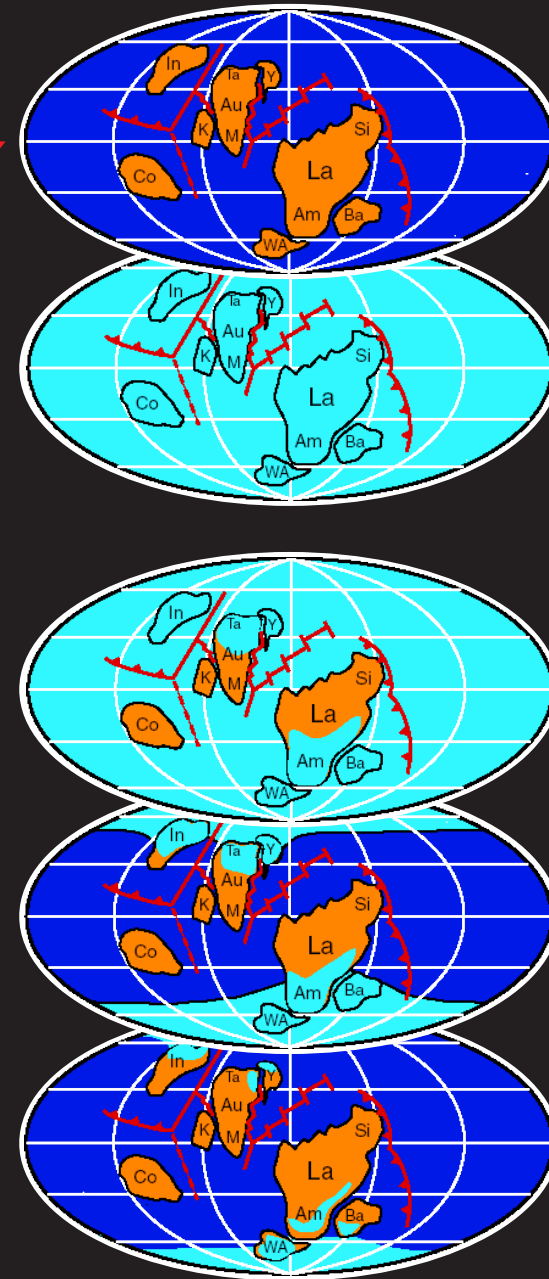
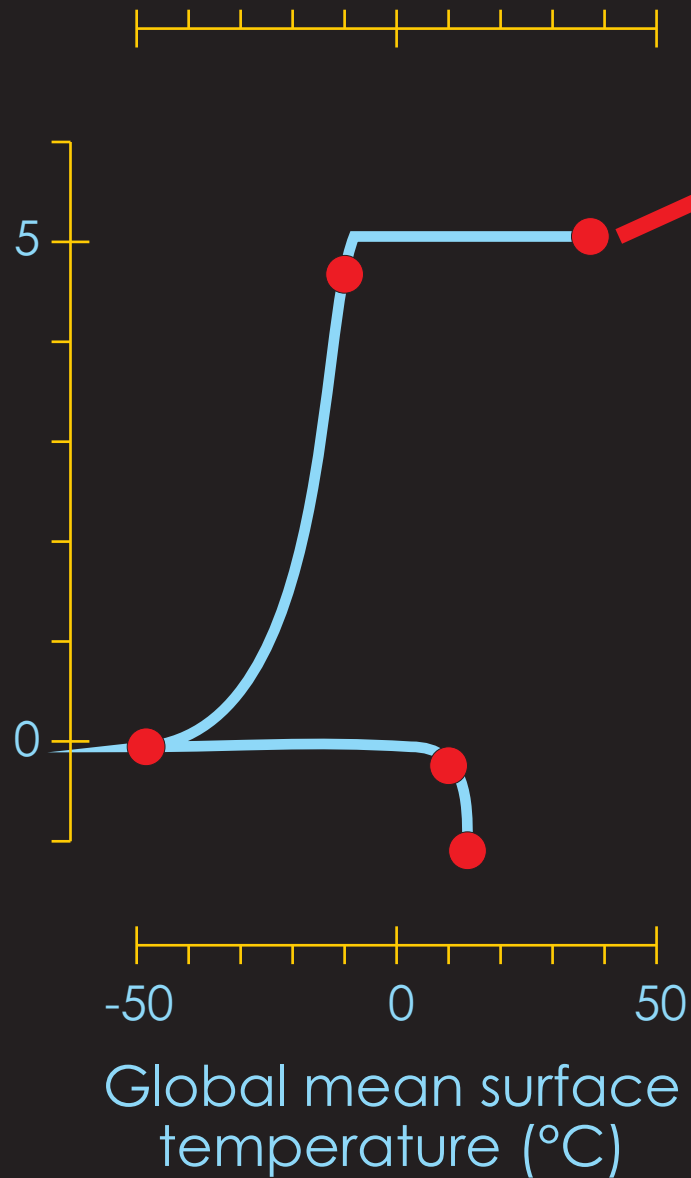
It can be seen that in (2) + (3), that the CO_2 removed (from the atmosphere) during weathering, is returned upon carbonate precipitation (and burial). In (1) + (3) (silicate weathering) CO_2 is permanently removed to the geological reservoir. This CO_2 must be balanced by mantle (/volcanic) out-gassing on the very long term.

The snowball Earth hypothesis

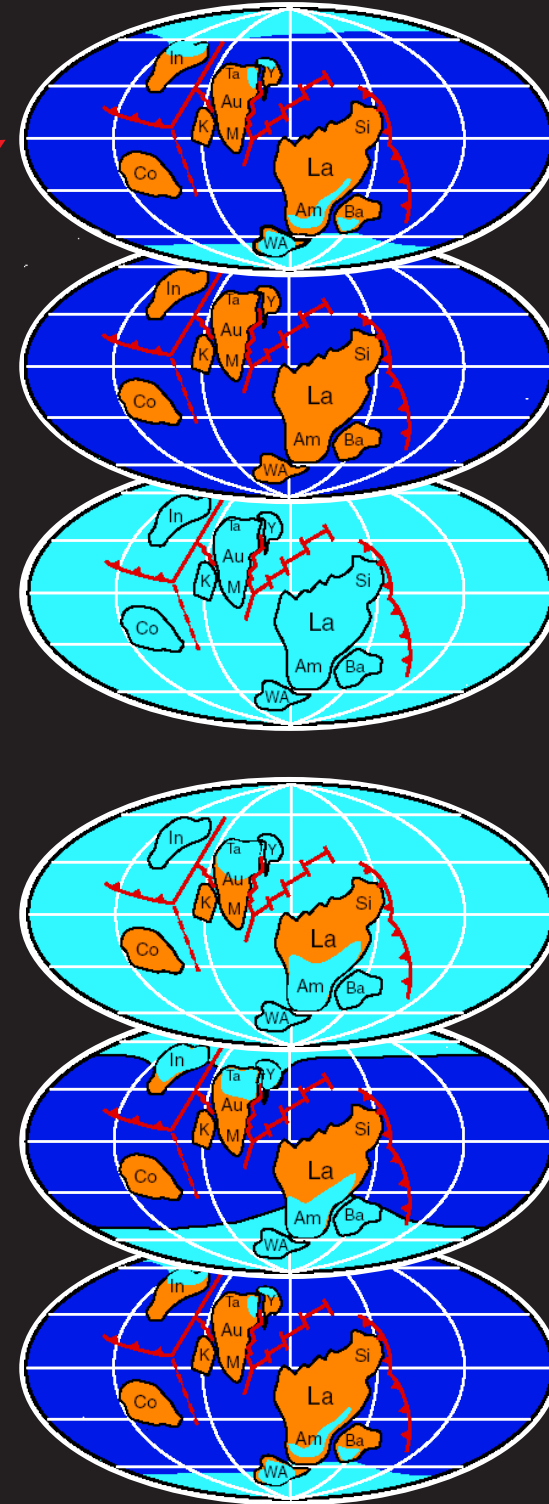
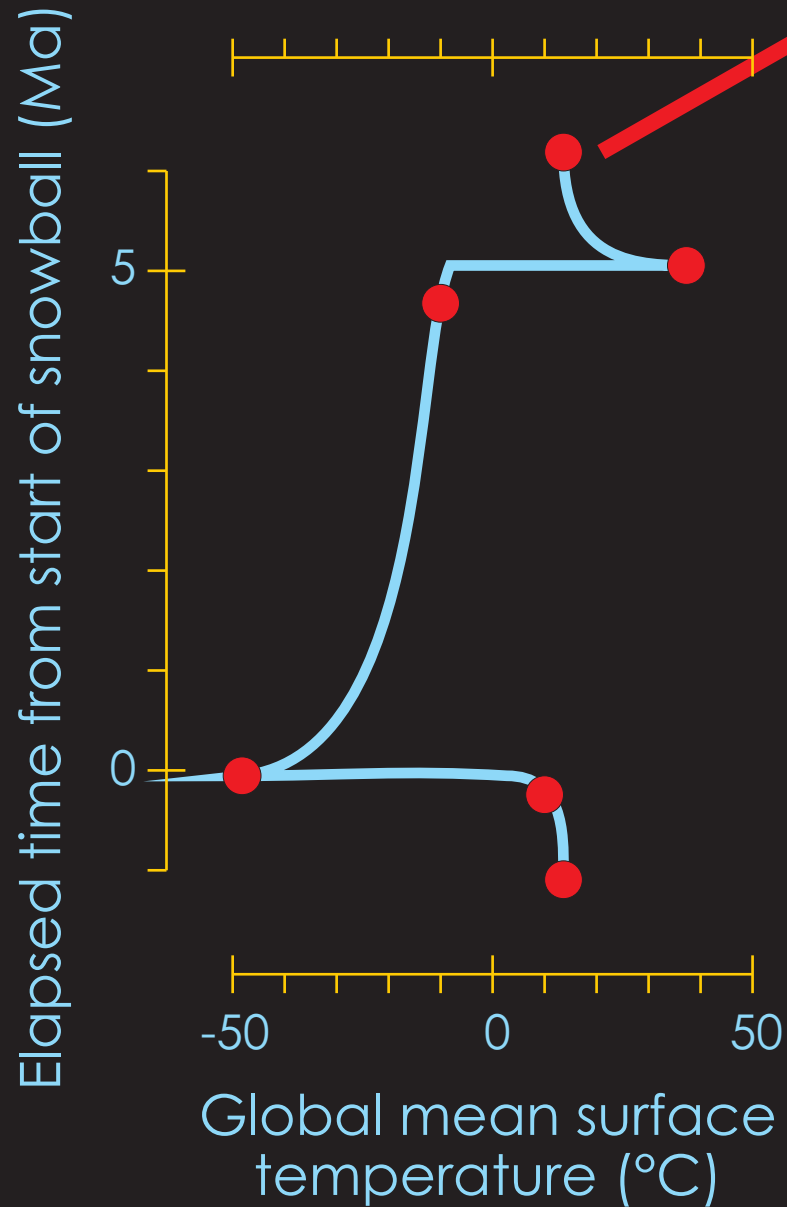


The snowball Earth hypothesis

Elapsed time from start of snowball (Ma)



The snowball Earth hypothesis

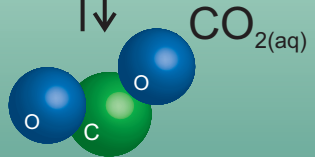
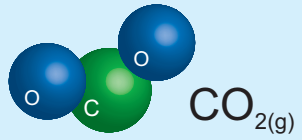


adapted from; Hoffman and Schrag [2002]

The enigma of the 'cap carbonates'



atmosphere



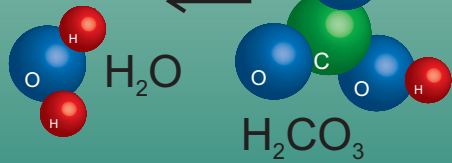
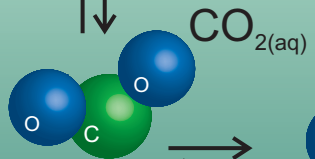
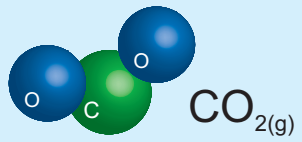
CO_2 chemistry
in seawater

ocean

From: *Barker and Ridgwell* [2012]

<http://www.nature.com/scitable/knowledge/library/ocean-acidification-25822734>

atmosphere

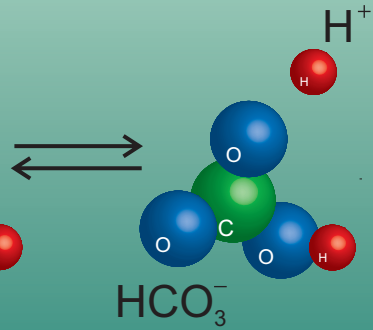
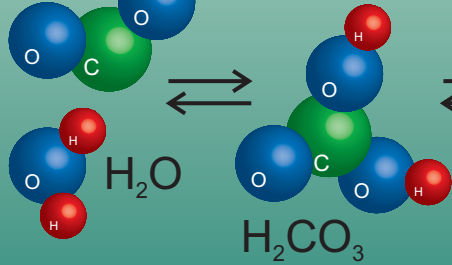
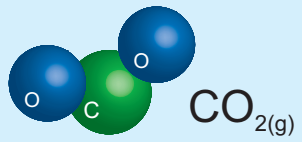


carbonic acid

ocean

CO_2 chemistry
in seawater

atmosphere

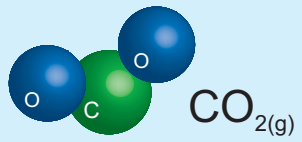


bicarbonate ion

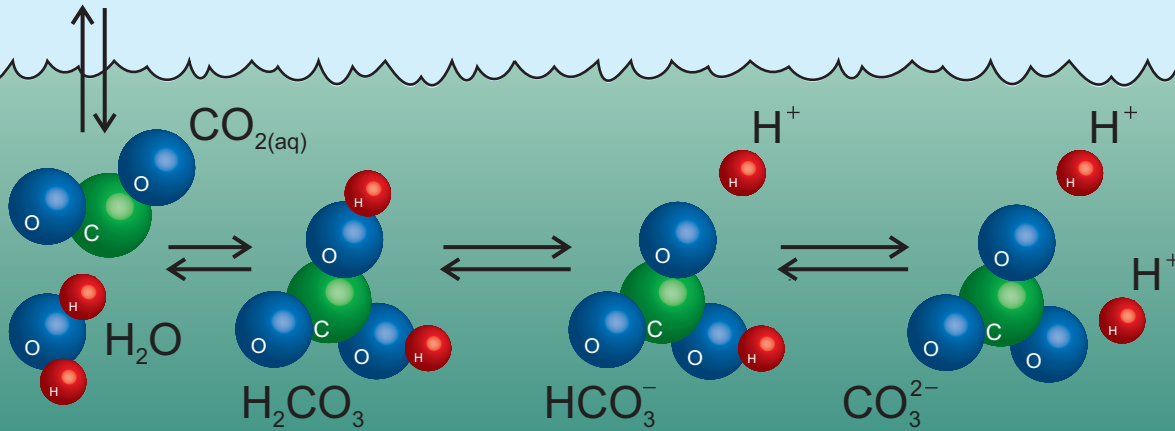
ocean

CO_2 chemistry
in seawater

atmosphere

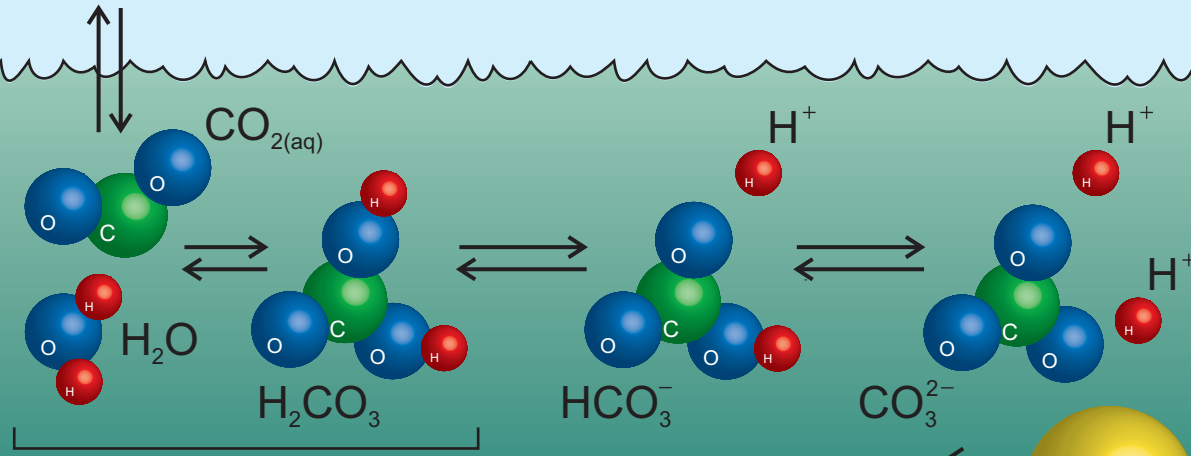
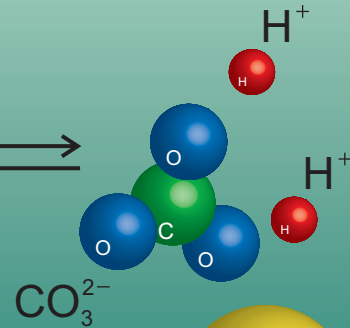
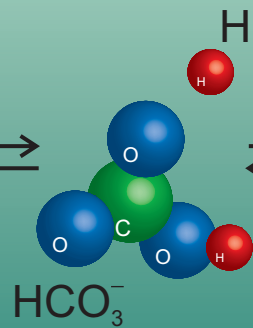
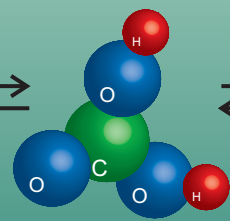
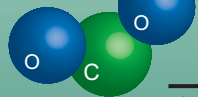
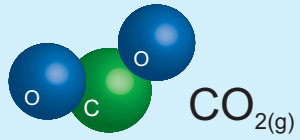


CO_2 chemistry
in seawater



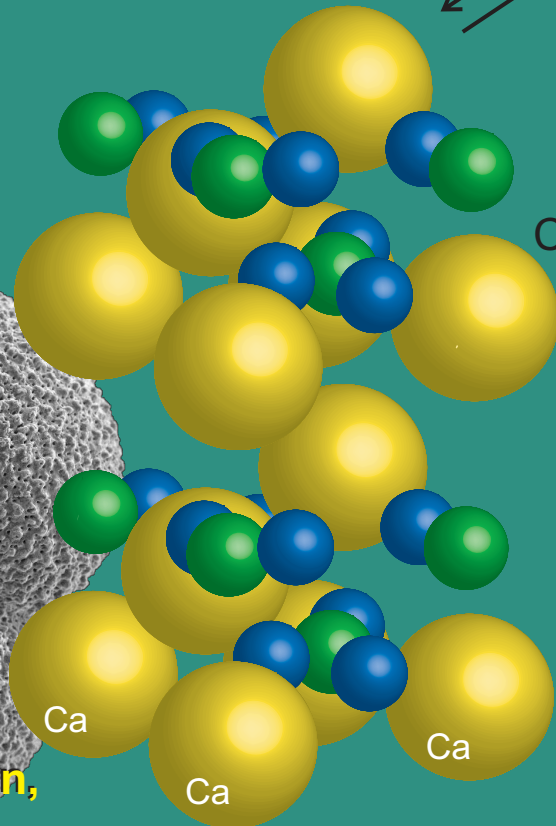
ocean

atmosphere



calcium carbonate mineral surface

(calcifying plankton, e.g. foraminifera)



ocean

CO_2 chemistry & mineral phases

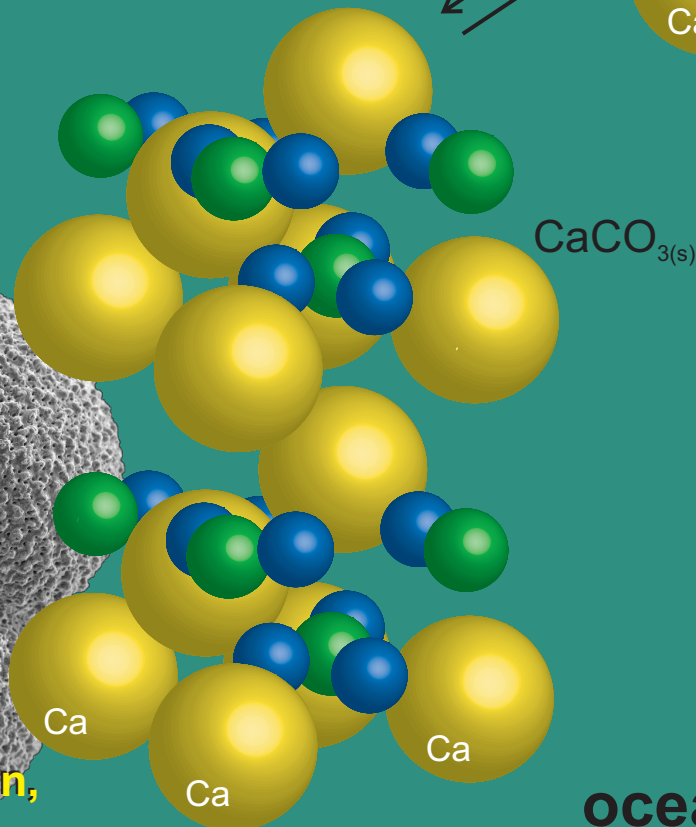
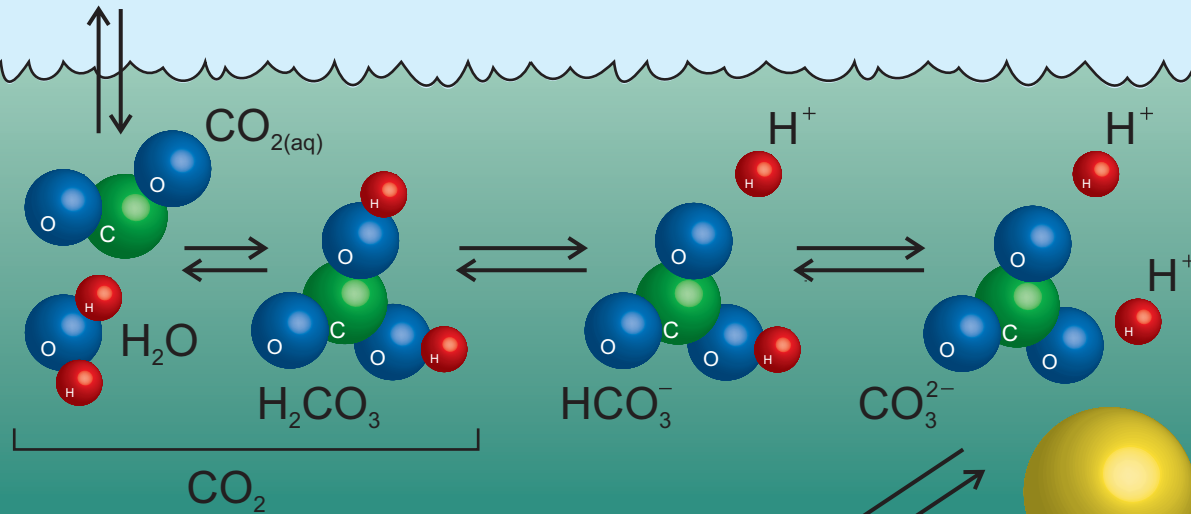
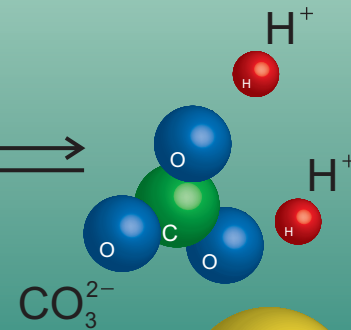
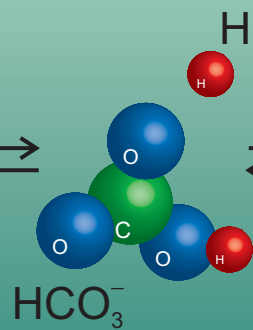
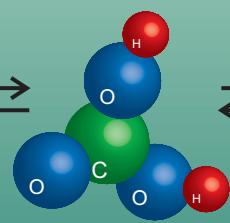
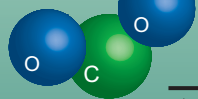
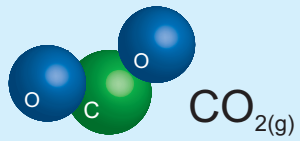


Aragonite: less stable
orthorhombic polymorph (e.g., many corals, pteropods)



Calcite: more stable
(and more abundant)
trigonal polymorph (e.g., coccolithophorides, foraminifera)

atmosphere



**calcium
carbonate
mineral
surface**

(calcifying plankton,
e.g. foraminifera)

ocean

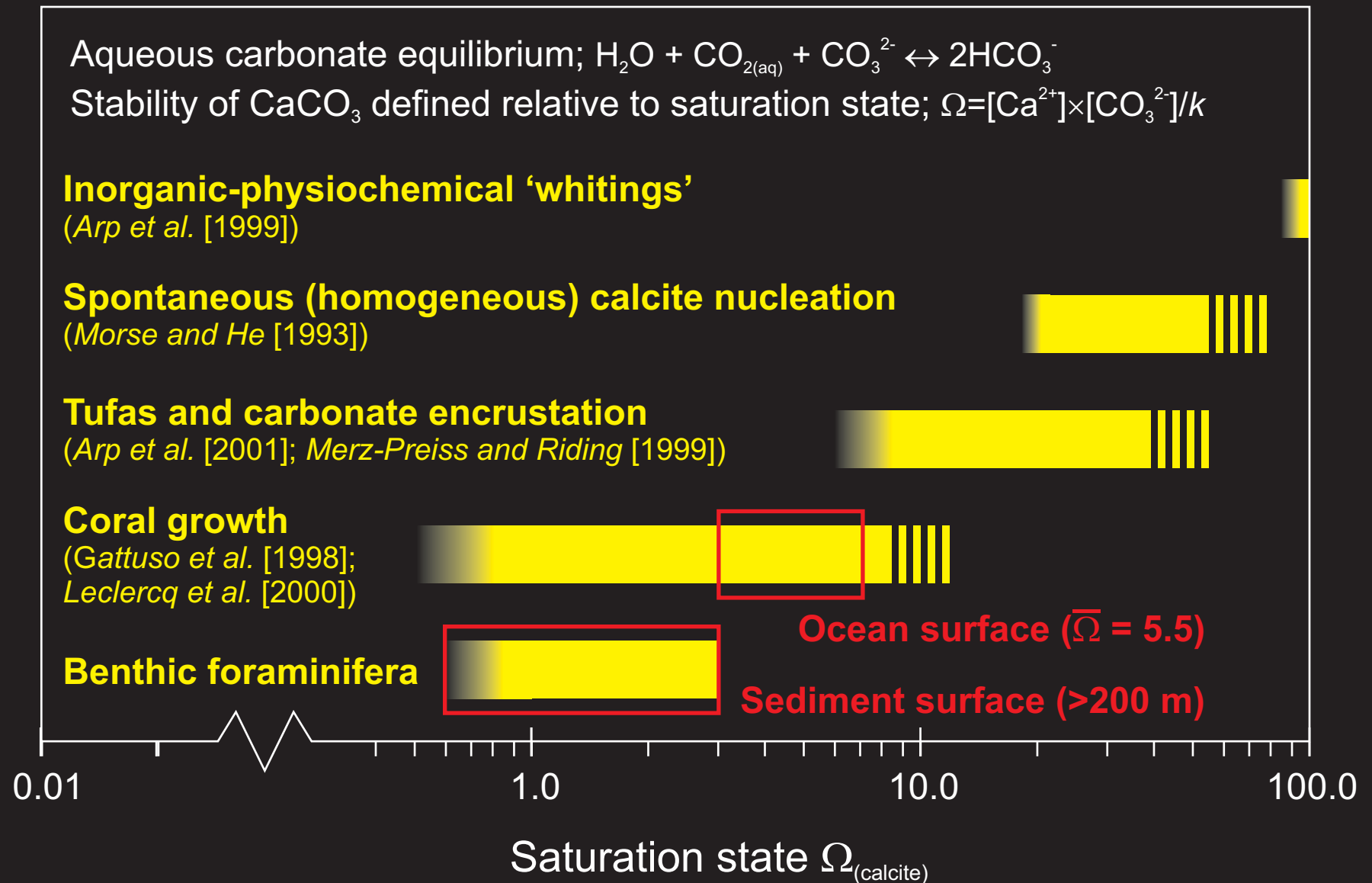
CO_2 chemistry & mineral phases

The addition of CO_2 to seawater results in a decrease in carbonate ion (CO_3^{2-}) concentration and 'ocean acidification'. A decrease in CO_3^{2-} , in turn, suppresses the stability of CaCO_3 , defined by its saturation state:

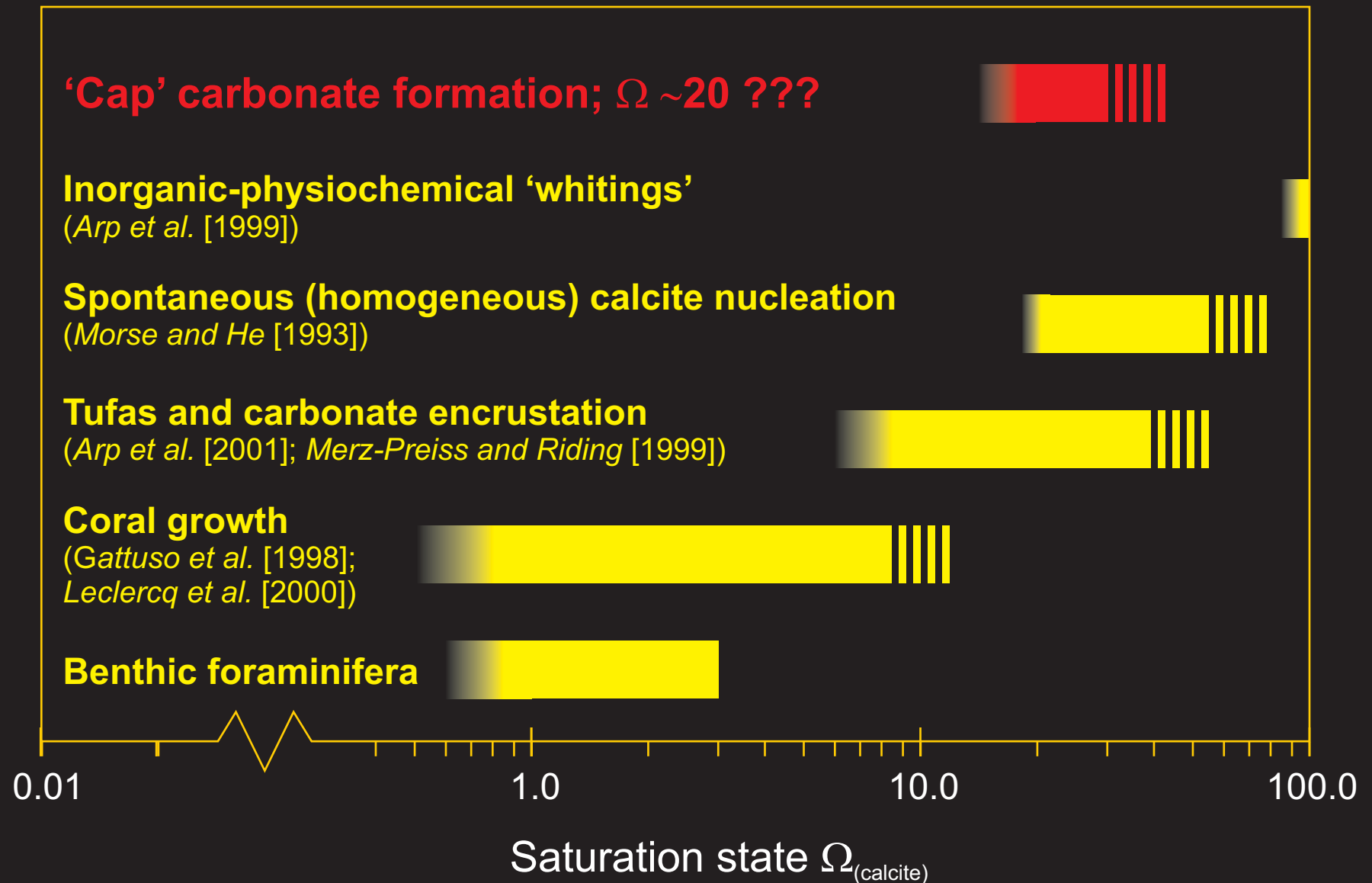
$$\Omega = [\text{Ca}^{2+}] \times [\text{CO}_3^{2-}] / k$$

\Rightarrow The thermodynamic efficiency of precipitating CaCO_3 is a function of $[\text{CO}_3^{2-}]$ (and carbonate 'saturation').

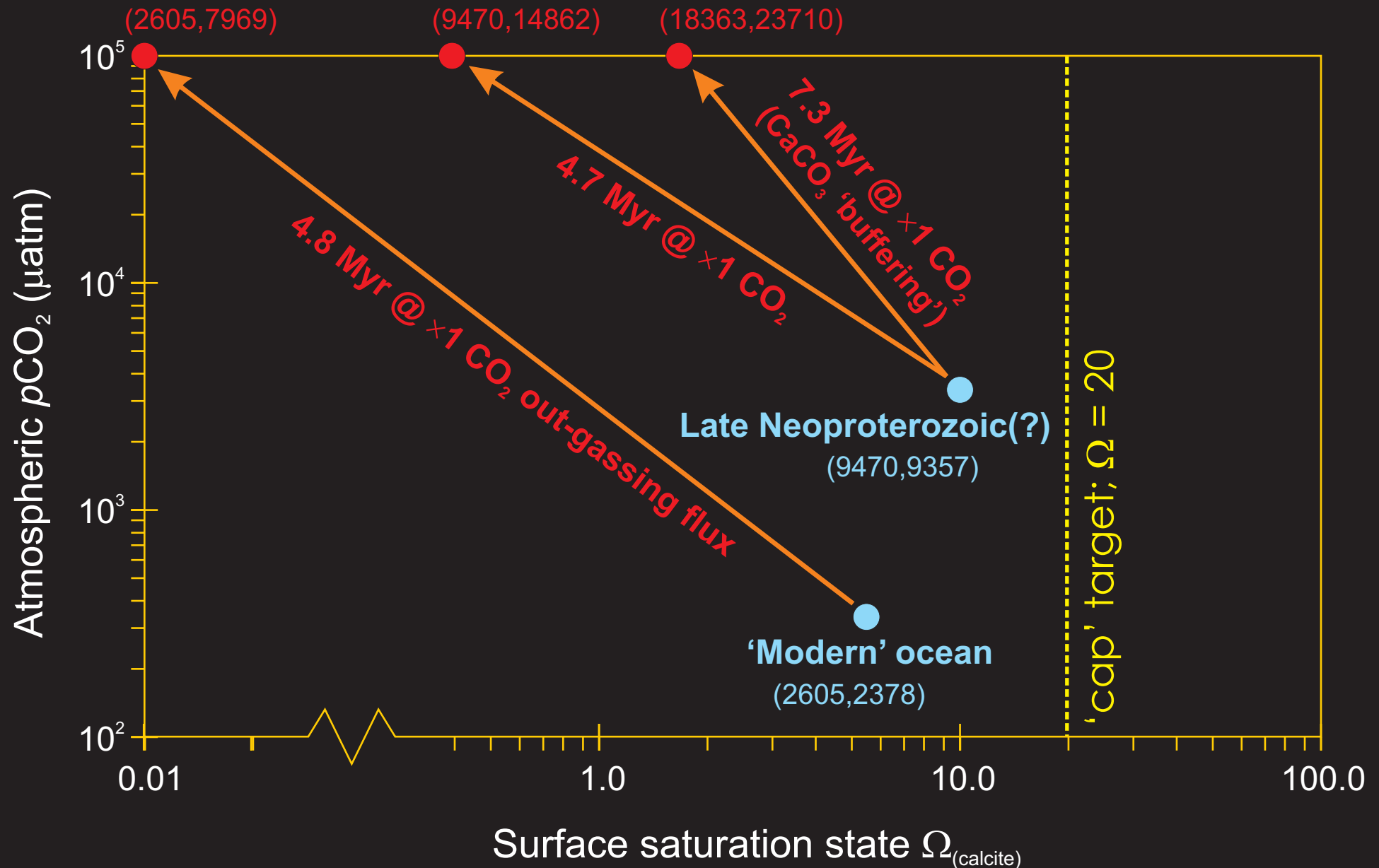
The enigma of the 'cap carbonates'



The enigma of the 'cap carbonates'

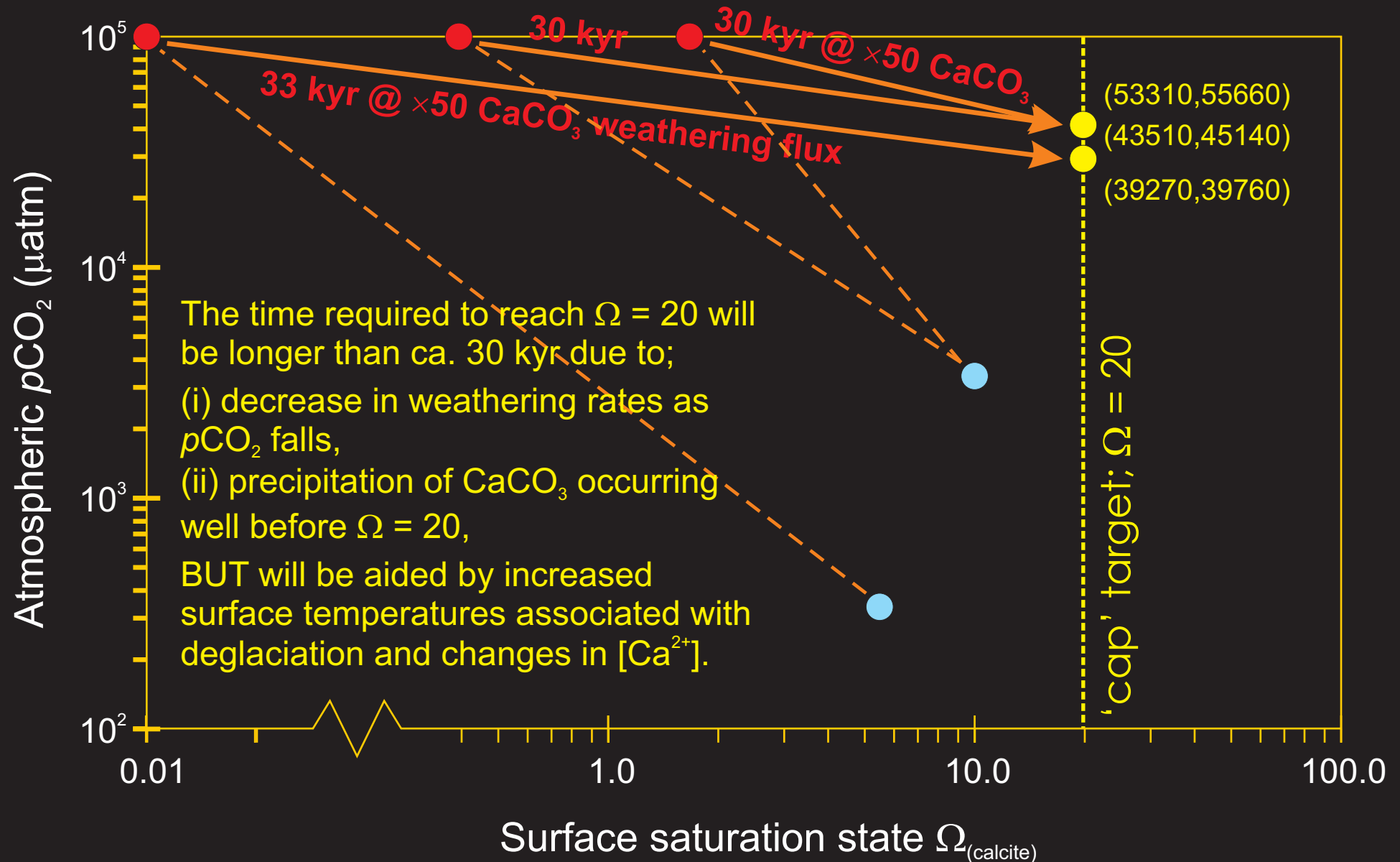


Potential evolution of ocean saturation during a 'snowball'



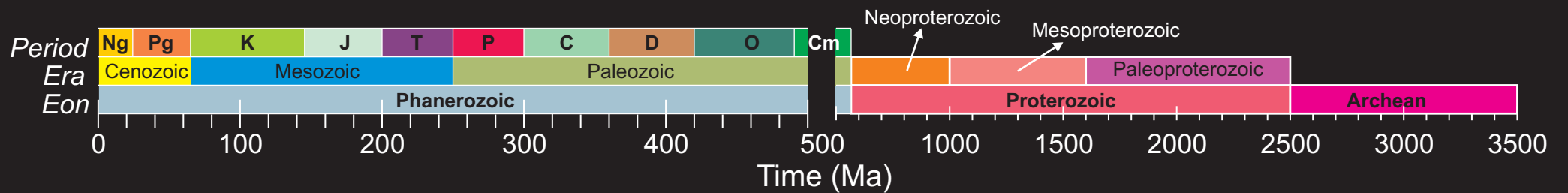
NOTE: ocean composition format;
[mean alkalinity, mean DIC] ($\mu\text{mol kg}^{-1}$)

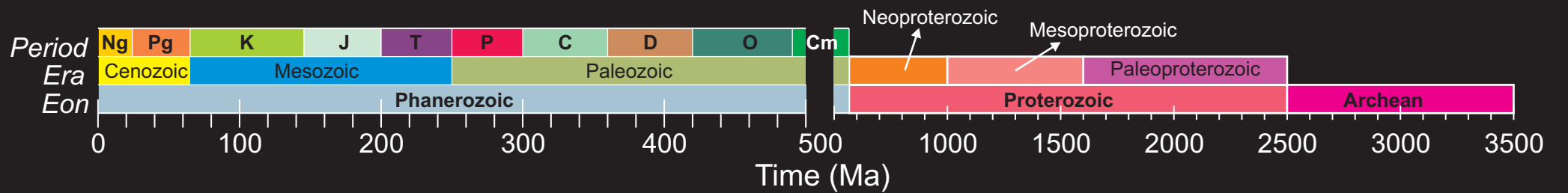
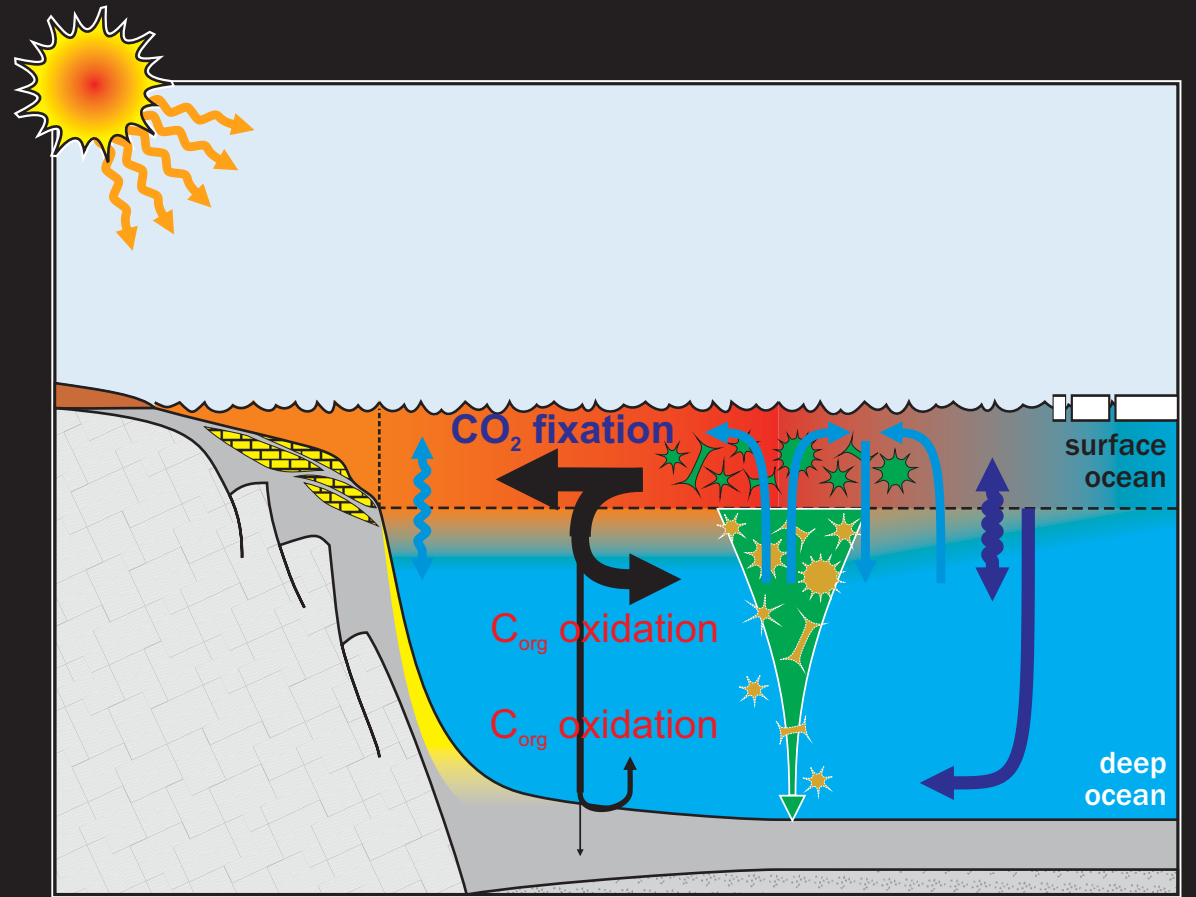
Potential evolution of ocean saturation during a 'snowball'

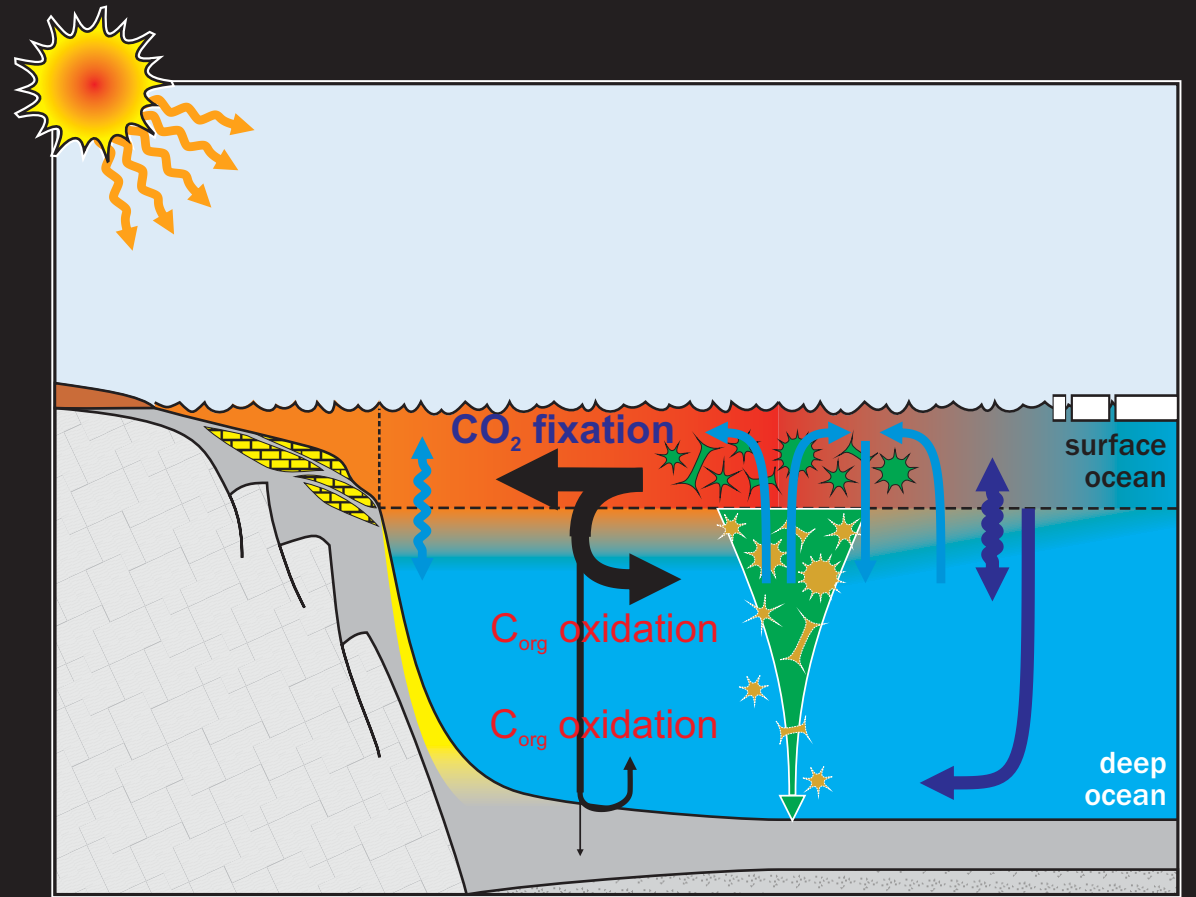


NOTE: ocean composition format;
[mean alkalinity, mean DIC] ($\mu\text{mol kg}^{-1}$)

Evolution of the Biological Pump

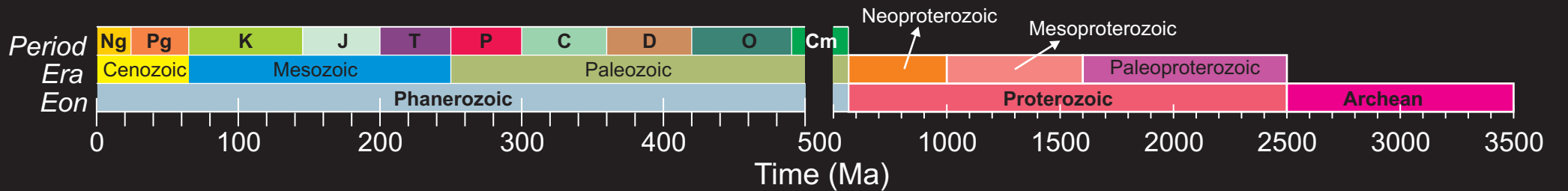






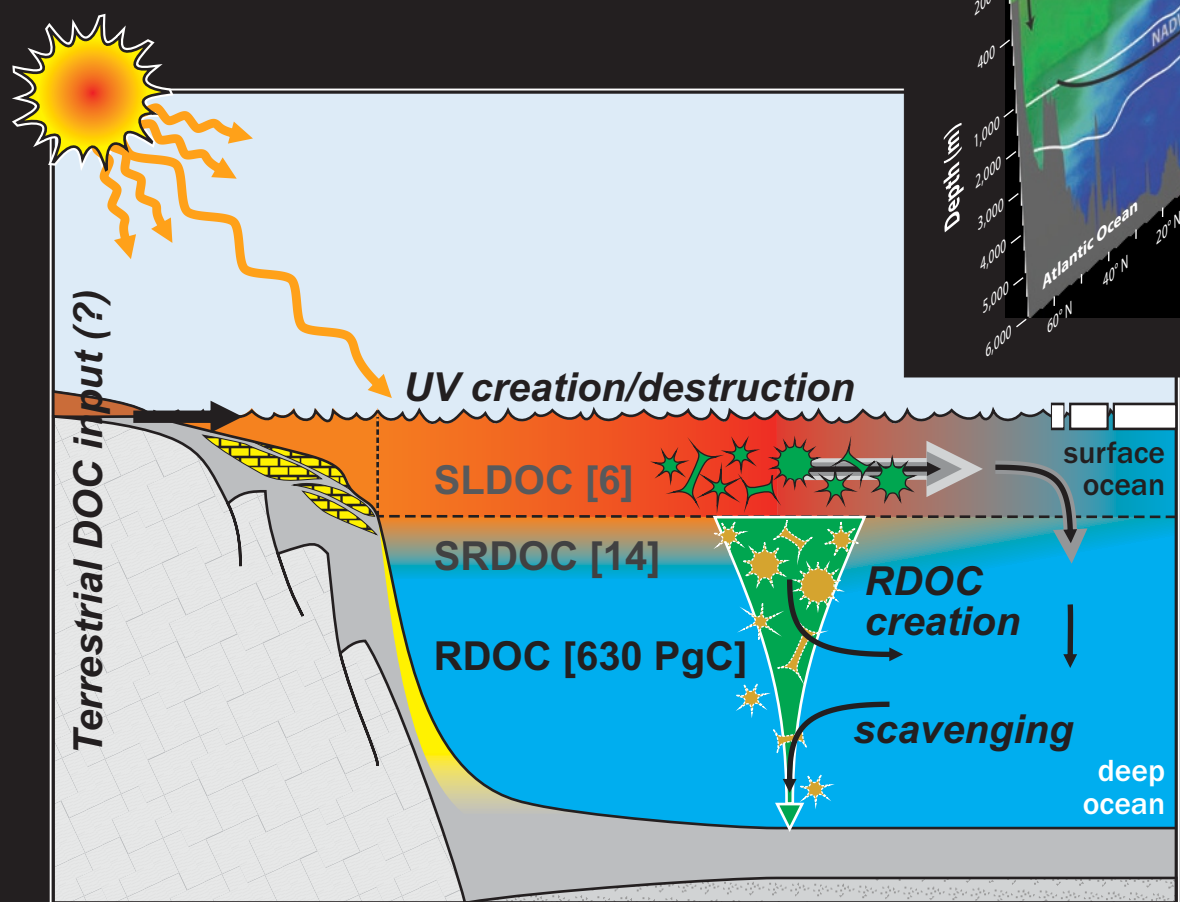
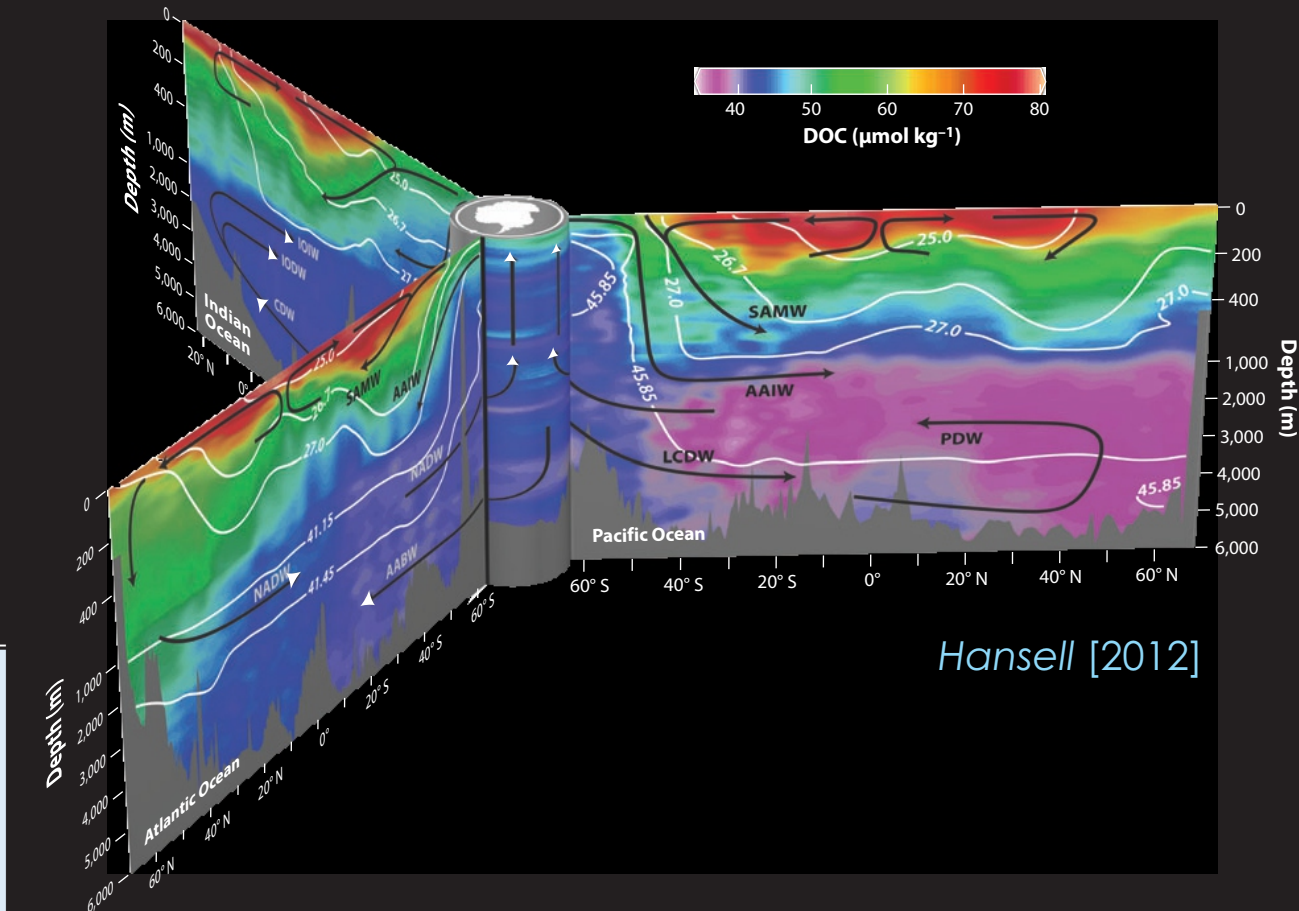
Major changes in plankton assemblage

Martin [1995]



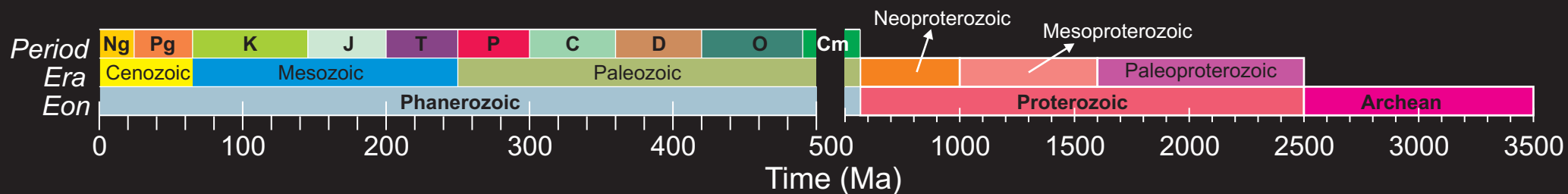
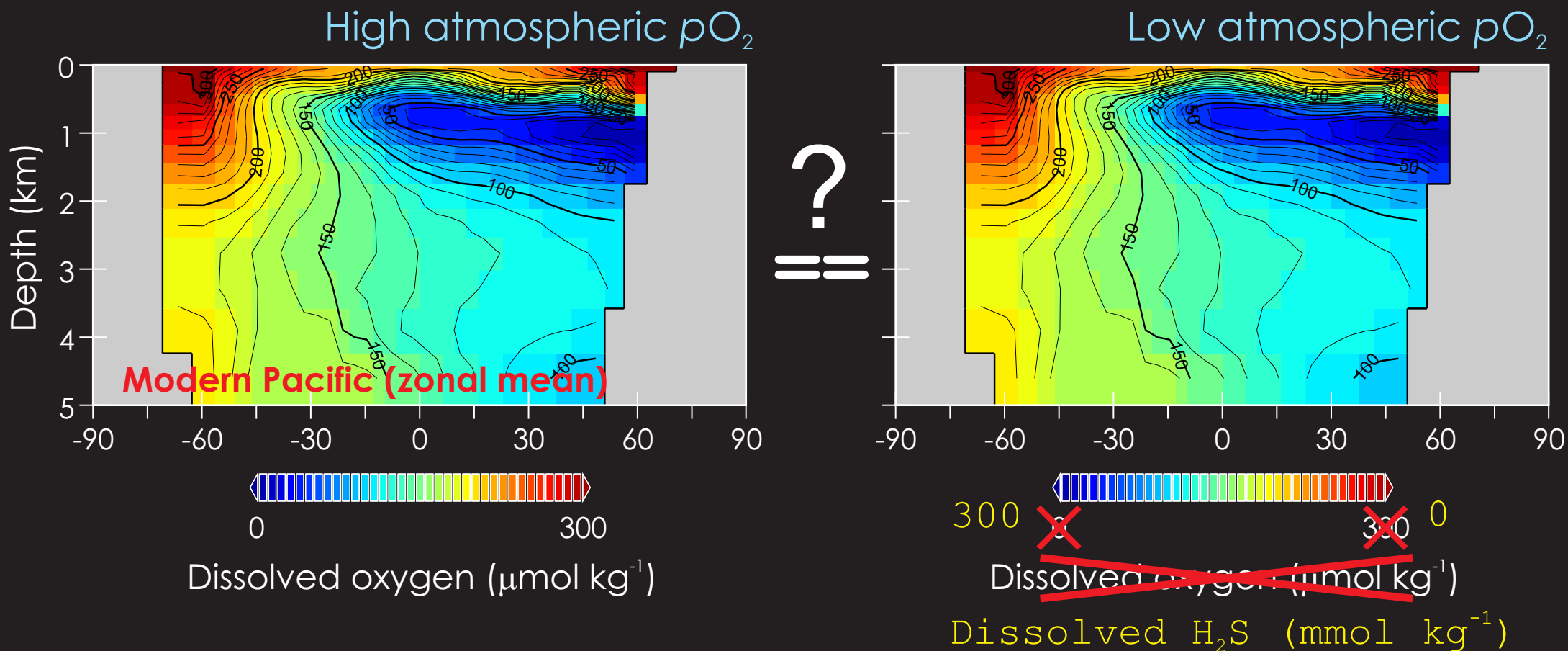
Evolution of the Biological Pump: Dissolved organic matter

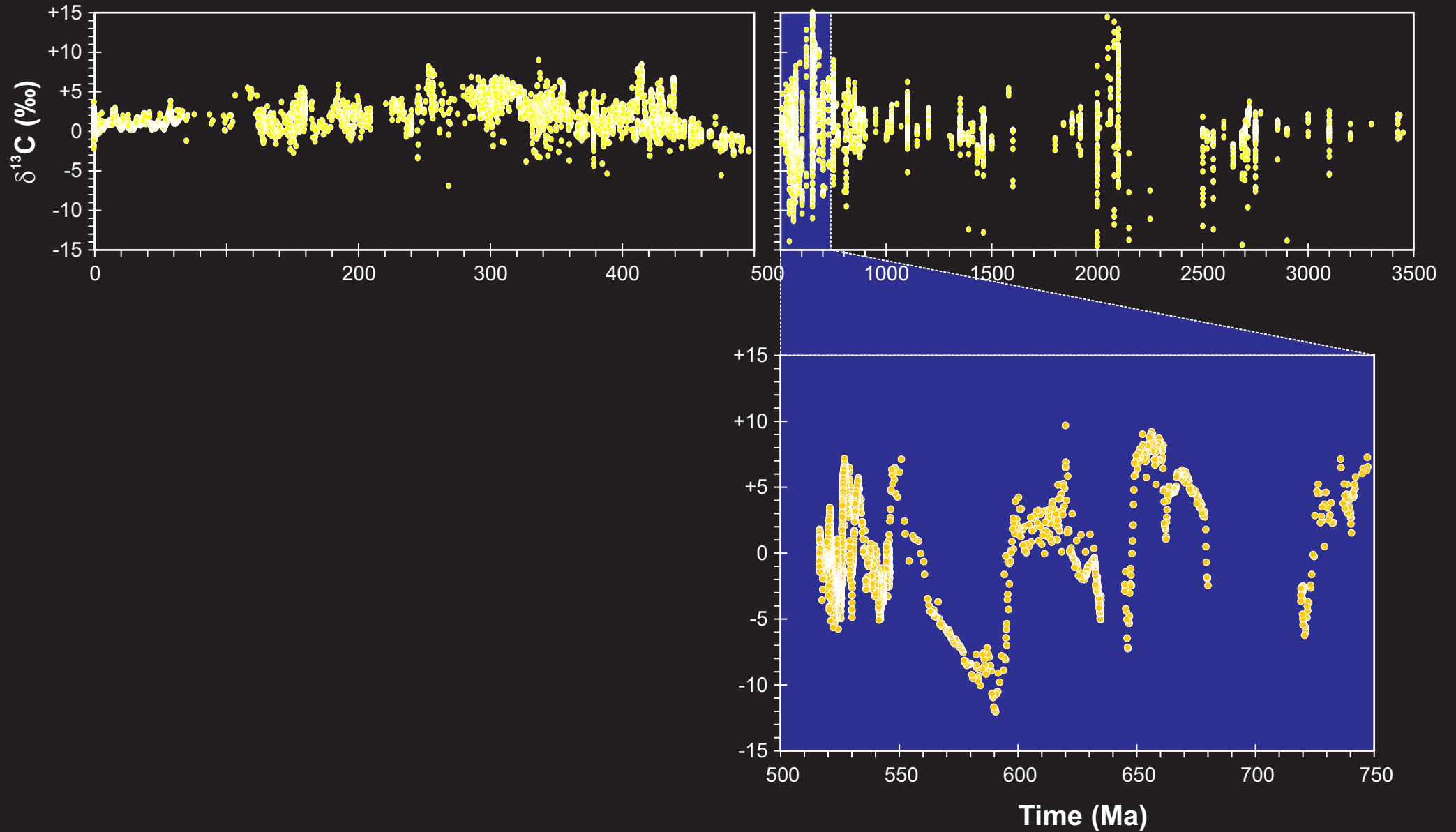
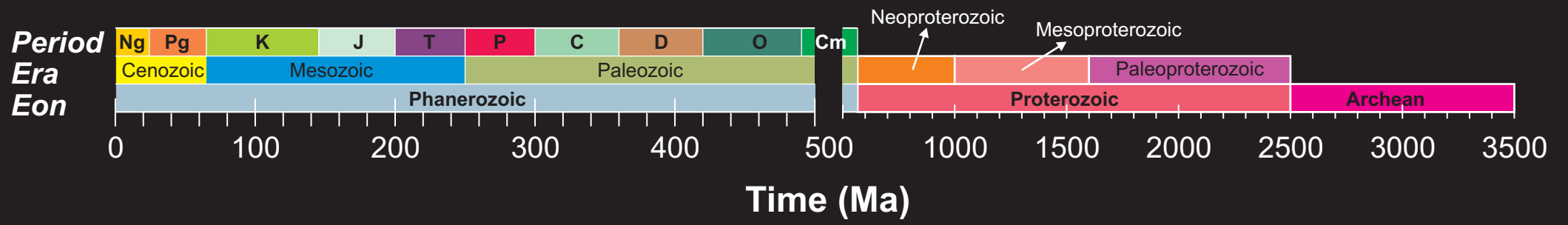
Dissolved organic matter



Evolution of the Biological Pump:

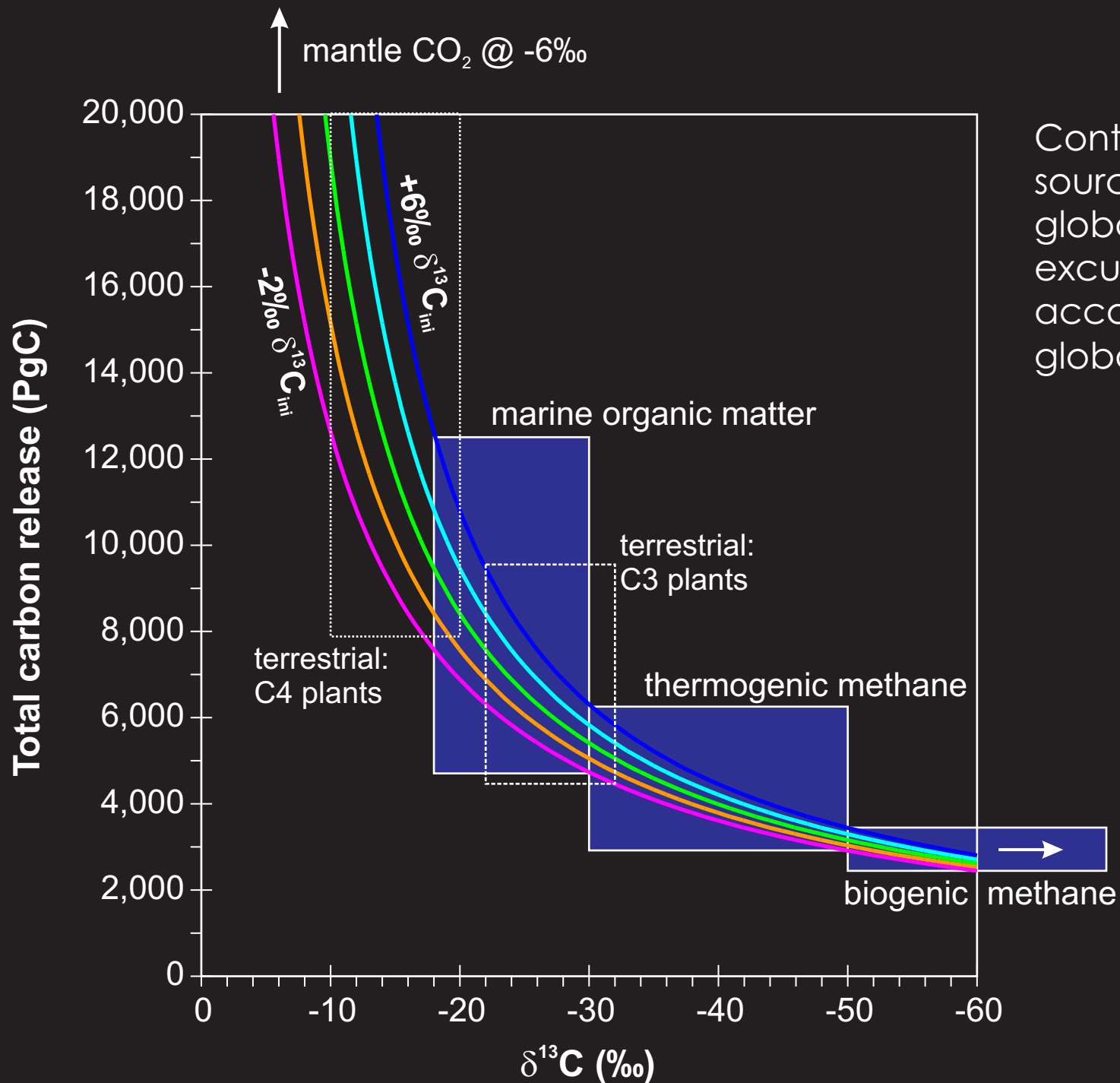
Dissolved organic matter





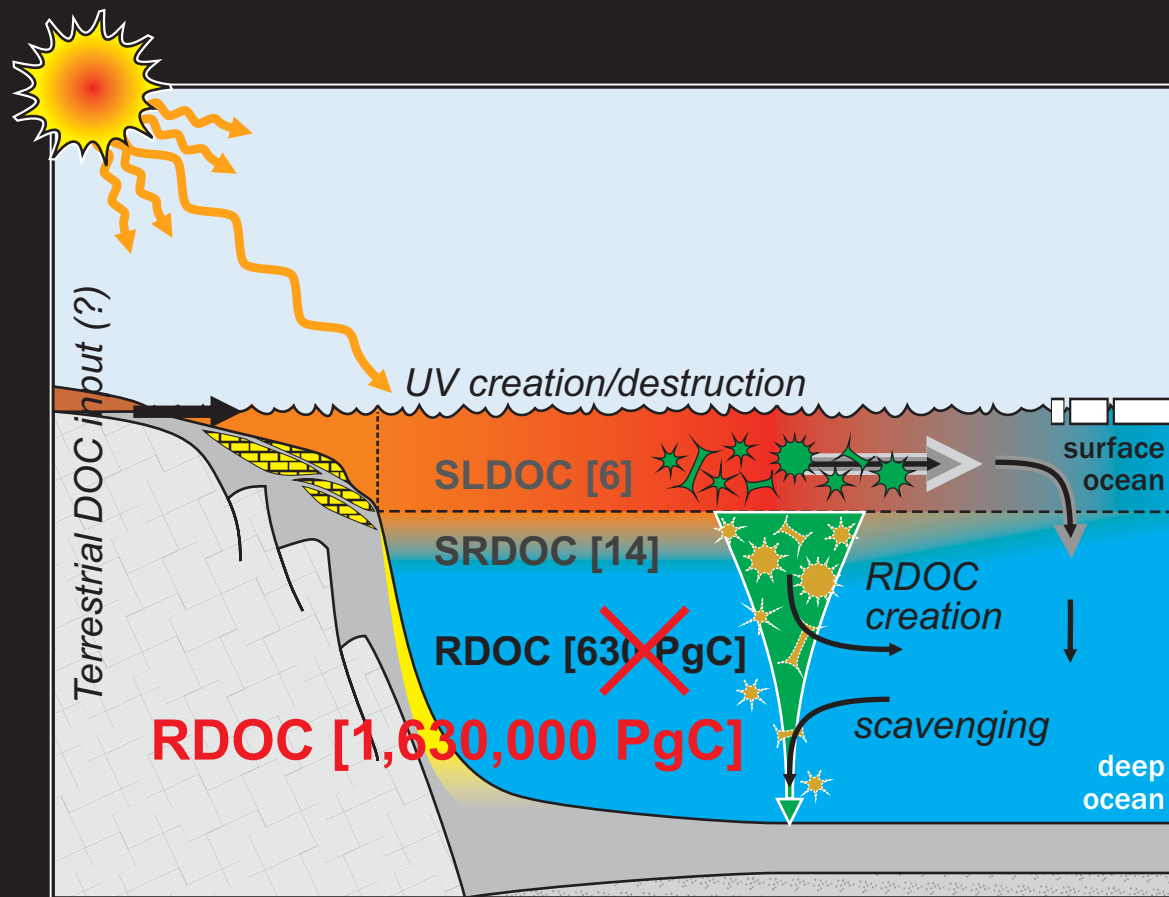
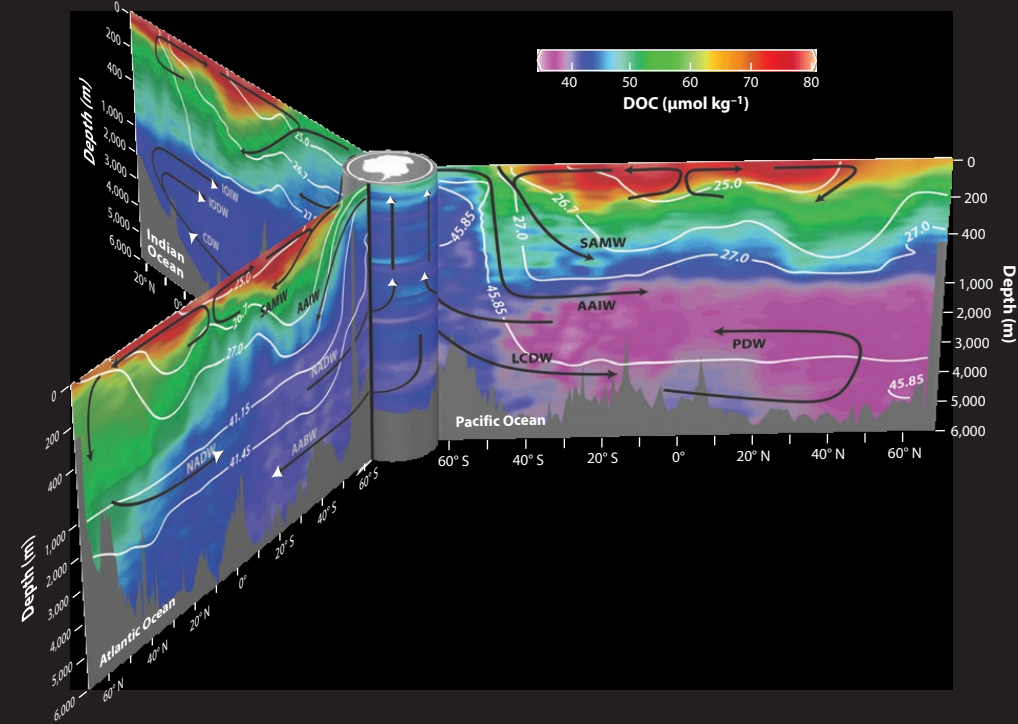
Evolution of the Biological Pump:

Dissolved organic matter

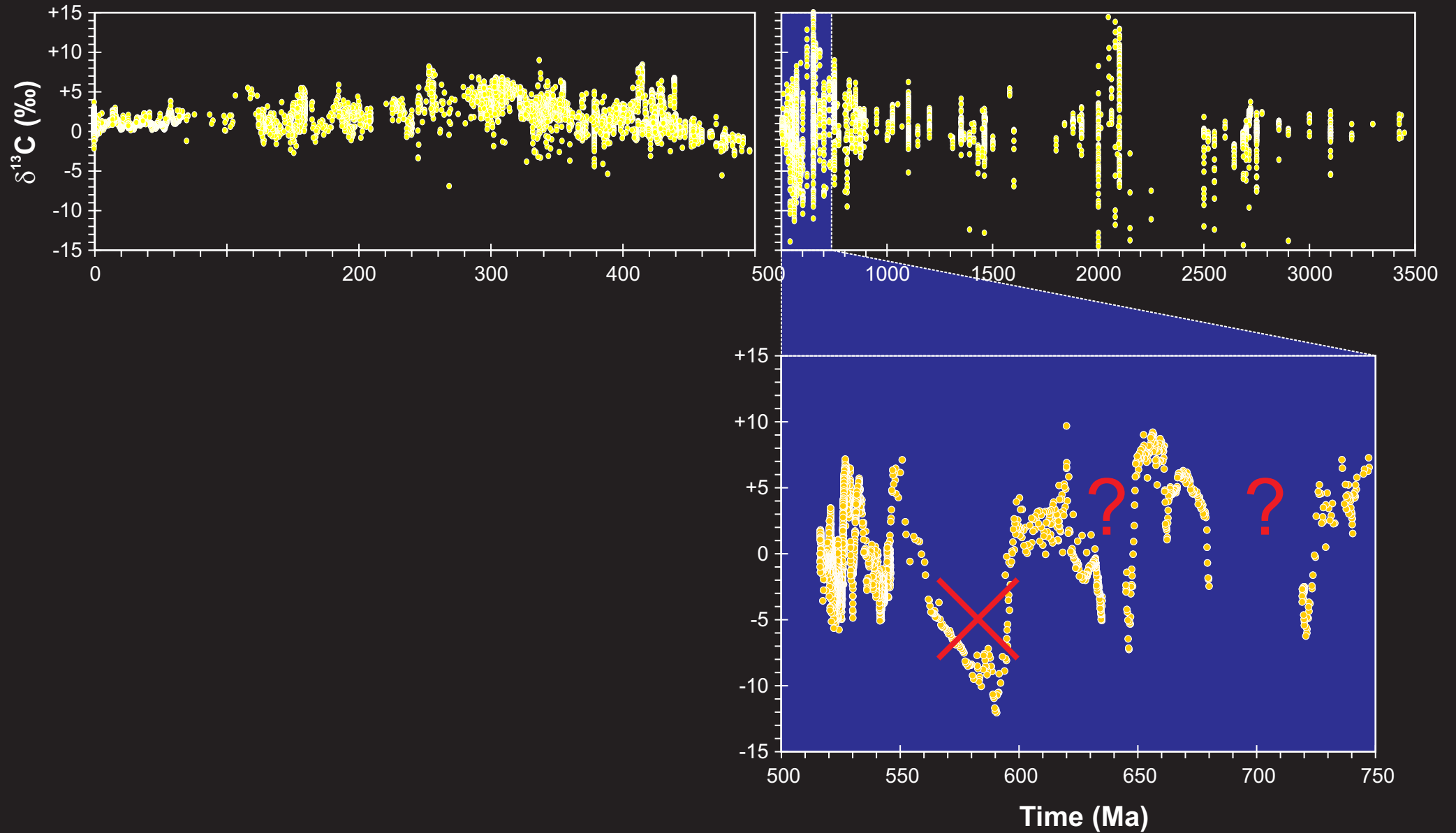
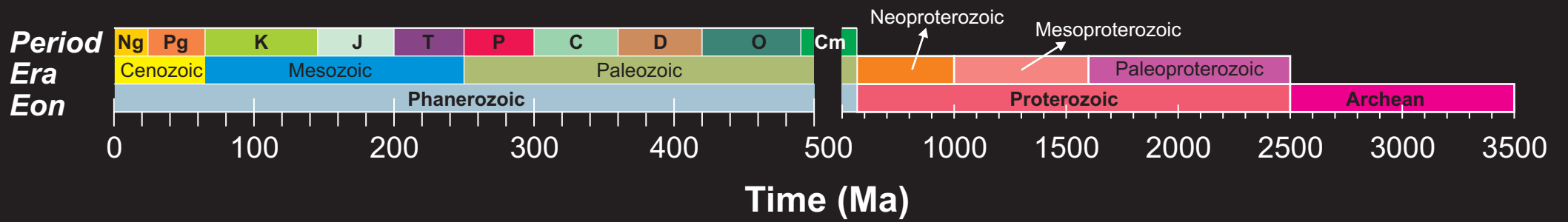


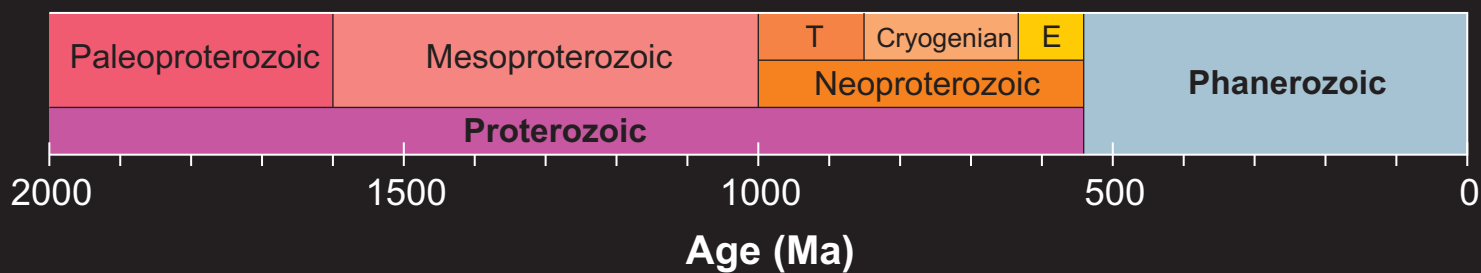
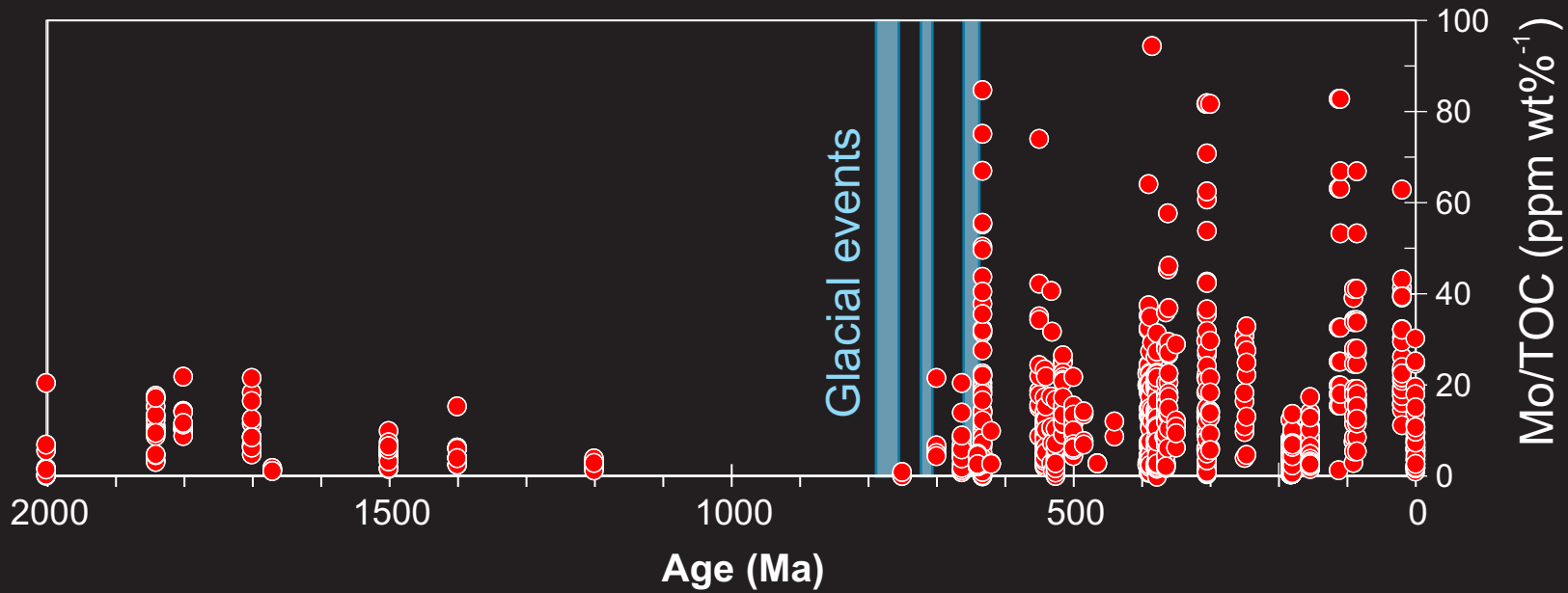
Contours of carbon release vs. source isotopic signature for a global -4‰ carbon isotopic excursion. Contours differ according to the initial mean global $\delta^{13}\text{C}$.

Evolution of the Biological Pump: Dissolved organic matter

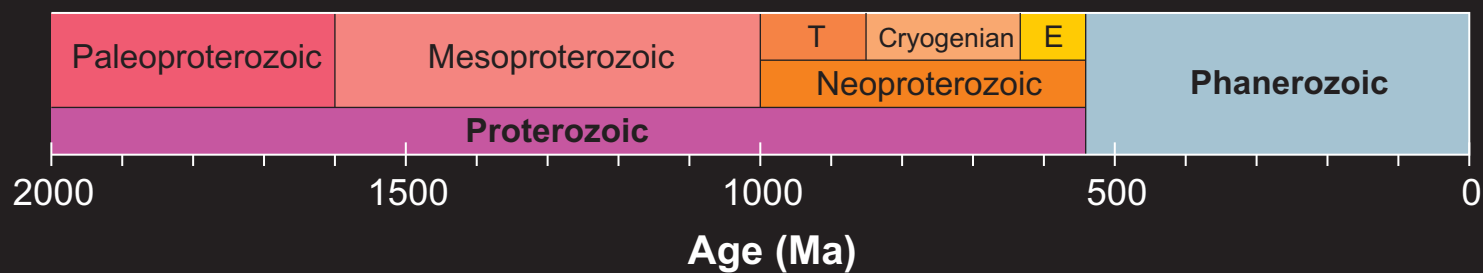
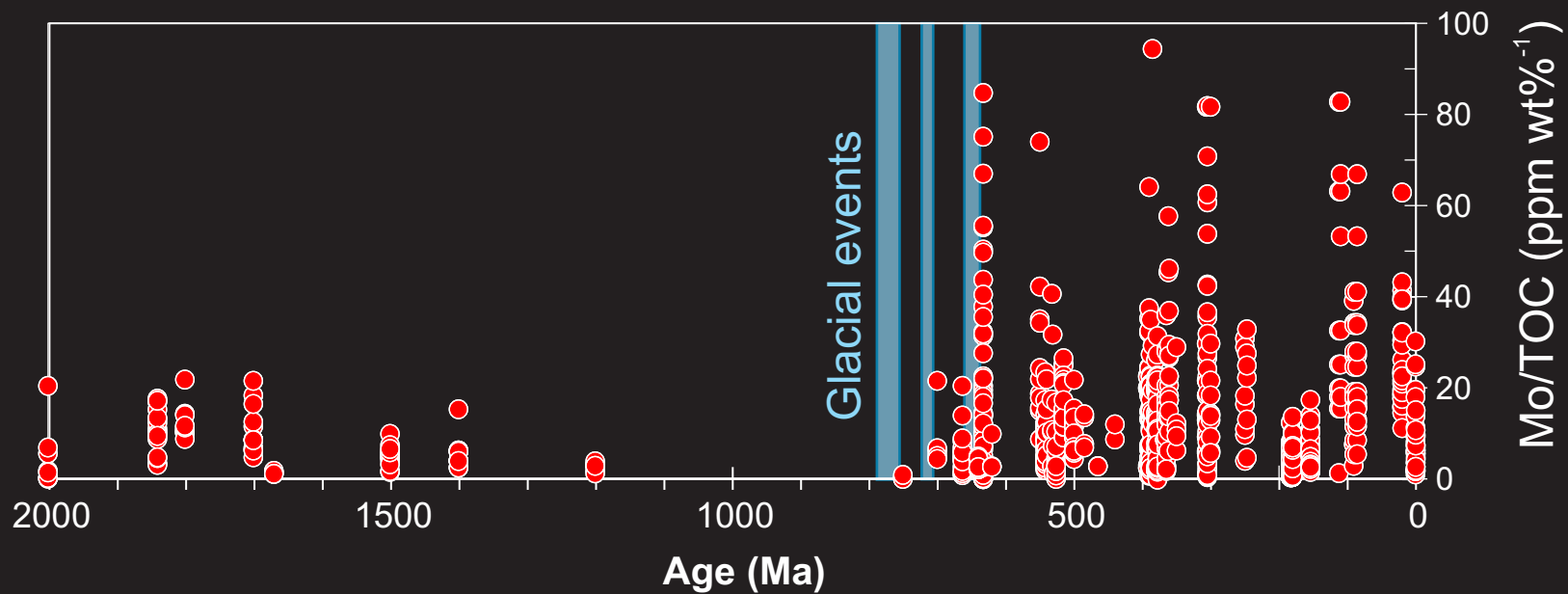


In the *Rothman et al.* [2003] model, the RDOC reservoir is assumed to have been at least 10 times the size of the inorganic (ocean DIC + atmospheric pCO_2) reservoir. For a modern DIC + pCO_2 reservoir of 39,000 PgC, this mean 390,000 PgC of DOC – more than 500 times larger than modern). For a higher late Precambrian DIC reservoir, the minimum DOC reservoir becomes 1.6×10^6 PgC, equivalent to concentration of a little over 1000 mgC per L of seawater and becoming the third most dominant dissolved species in the ocean after Cl^- .





Low fixed N supply to the open ocean
Low open ocean primary production



Low fixed N supply to the open ocean
Low open ocean primary production

Transitional interval

High diversity of N fixers
High primary production

